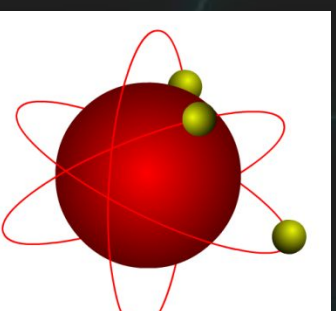
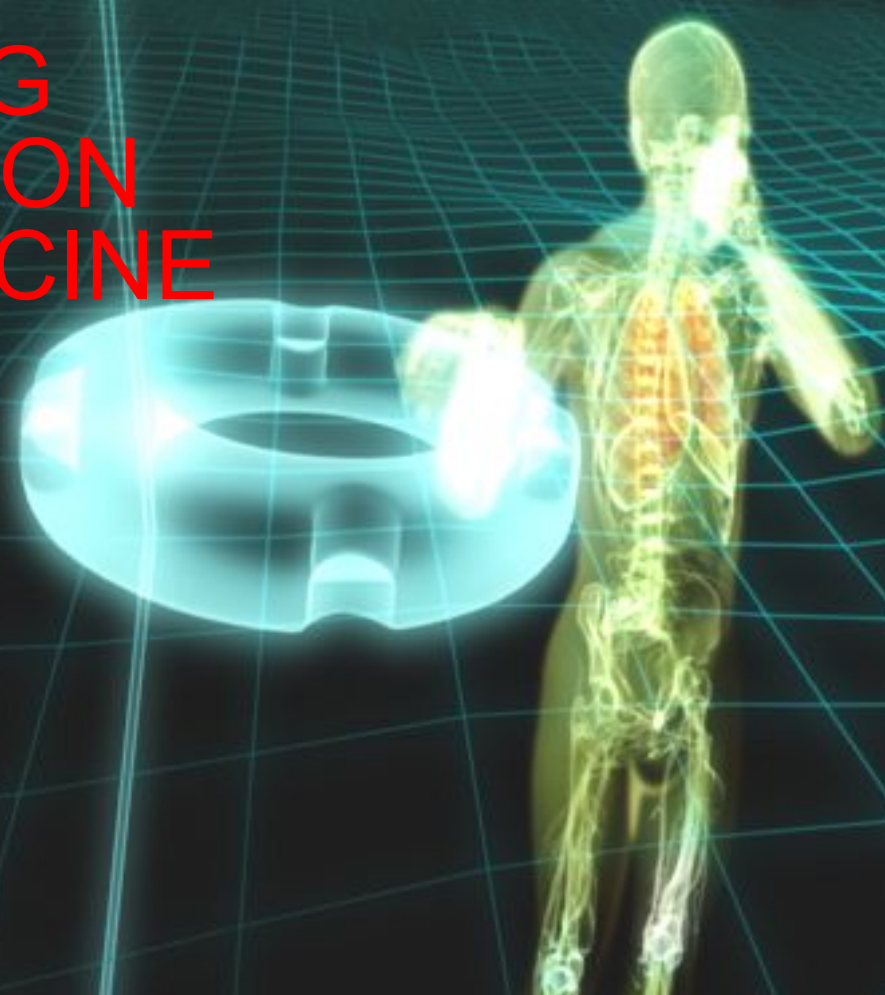
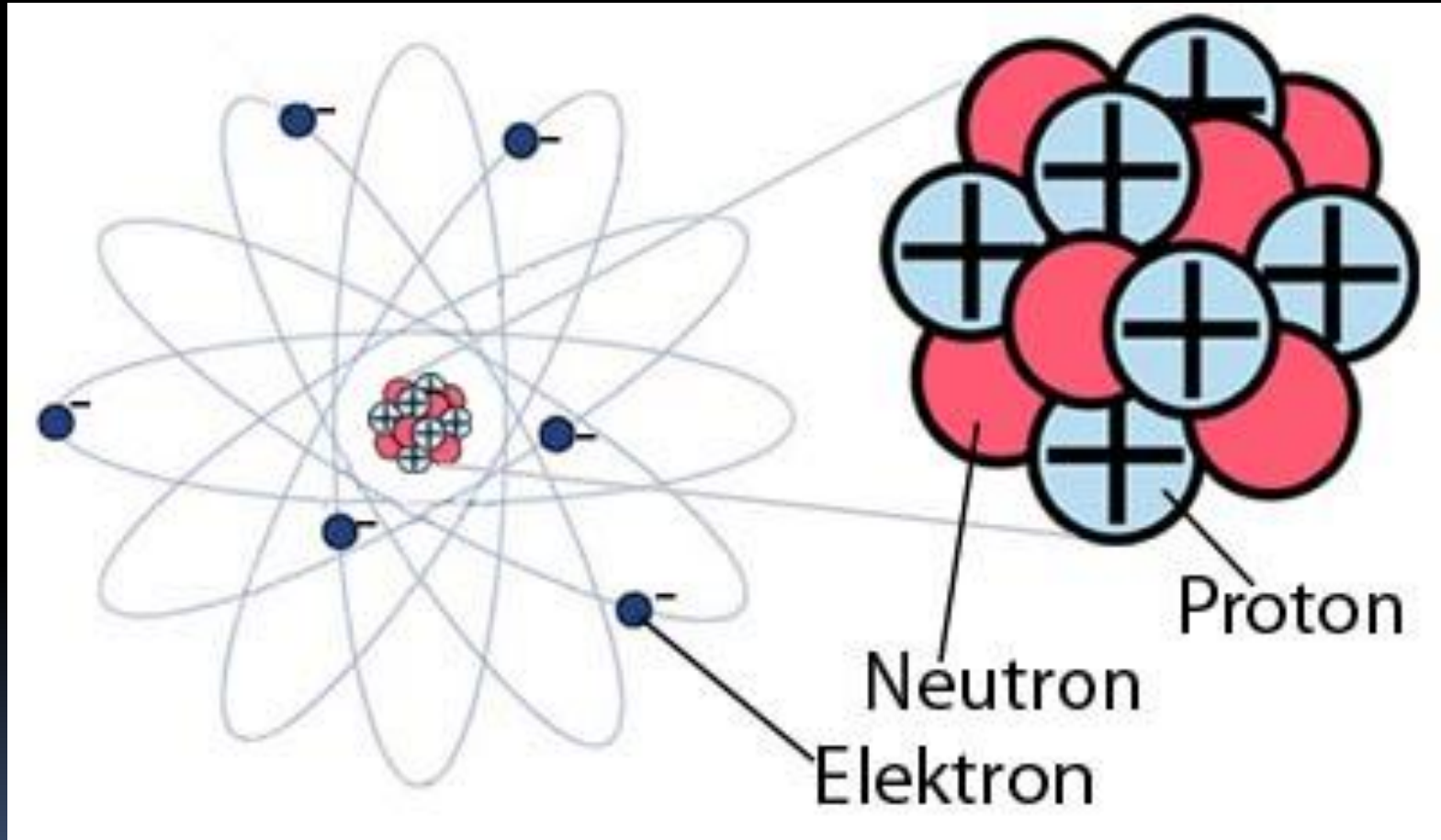




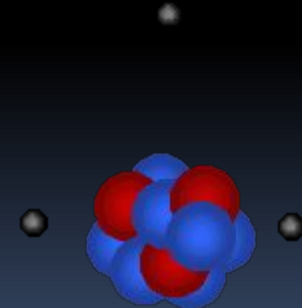
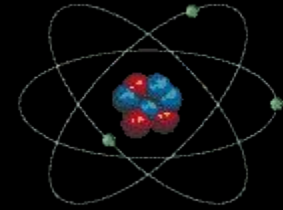
IONIZING RADIATION IN MEDICINE



NUCLEUS. ATOM.



NUCLEUS. ATOM.

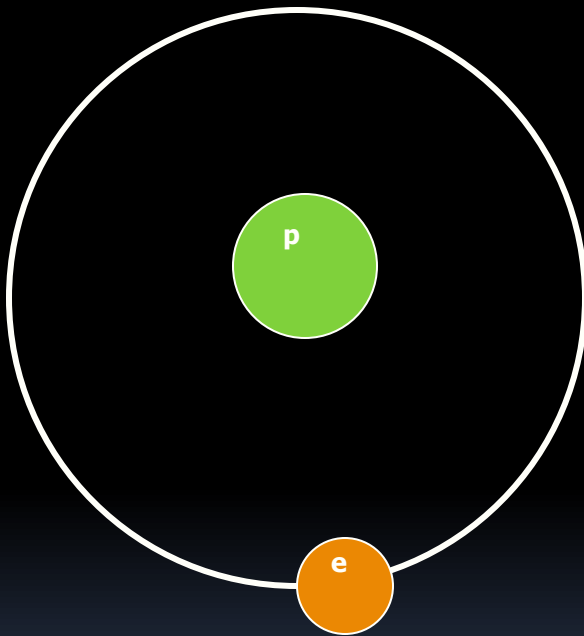


Parts of an Atom

- Each element in the Periodic Table has a different number of protons in its nucleus
 - Protons have positive charge
 - Change the number of protons □ change elements
 - This is called nuclear physics
- The element also has the same number of electrons
 - Electrons have negative charge
 - Change the number of electrons □ ionize the element
 - This is called chemistry
- Some elements also have neutrons
 - Neutrons have no charge
 - They are in the nuclei of atoms

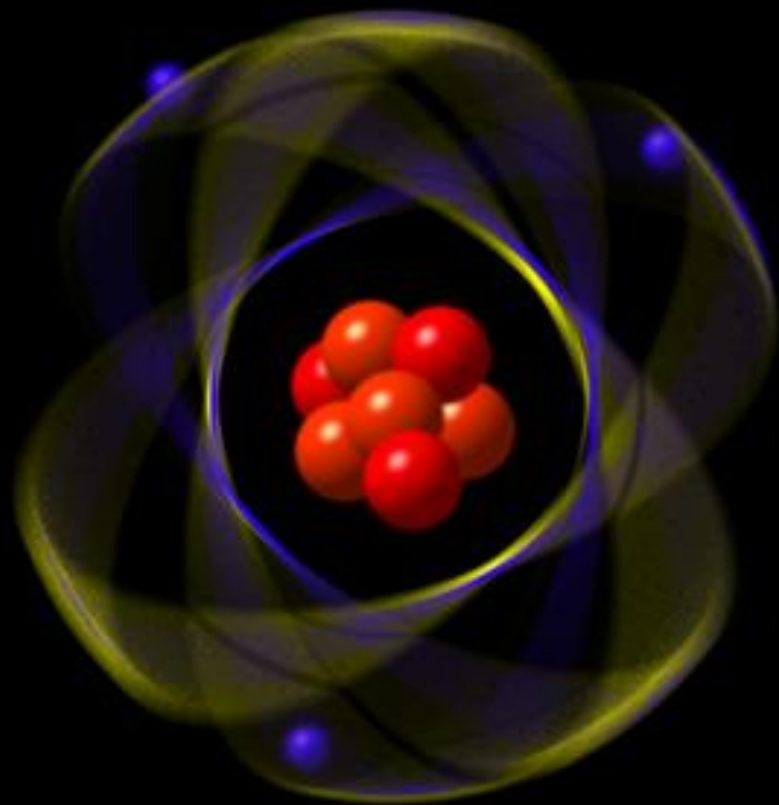


The Hydrogen Atom



${}^1\text{H}$

- One electron orbiting a nucleus
- 1 proton = Z = atomic number
- 0 neutrons = N
- Total mass = $A = Z+N = 1$
- Singly ionized Hydrogen is missing one electron = ${}^1\text{H}^+$
- Add a neutron and you have Deuterium = ${}^2\text{H} = \text{D}$



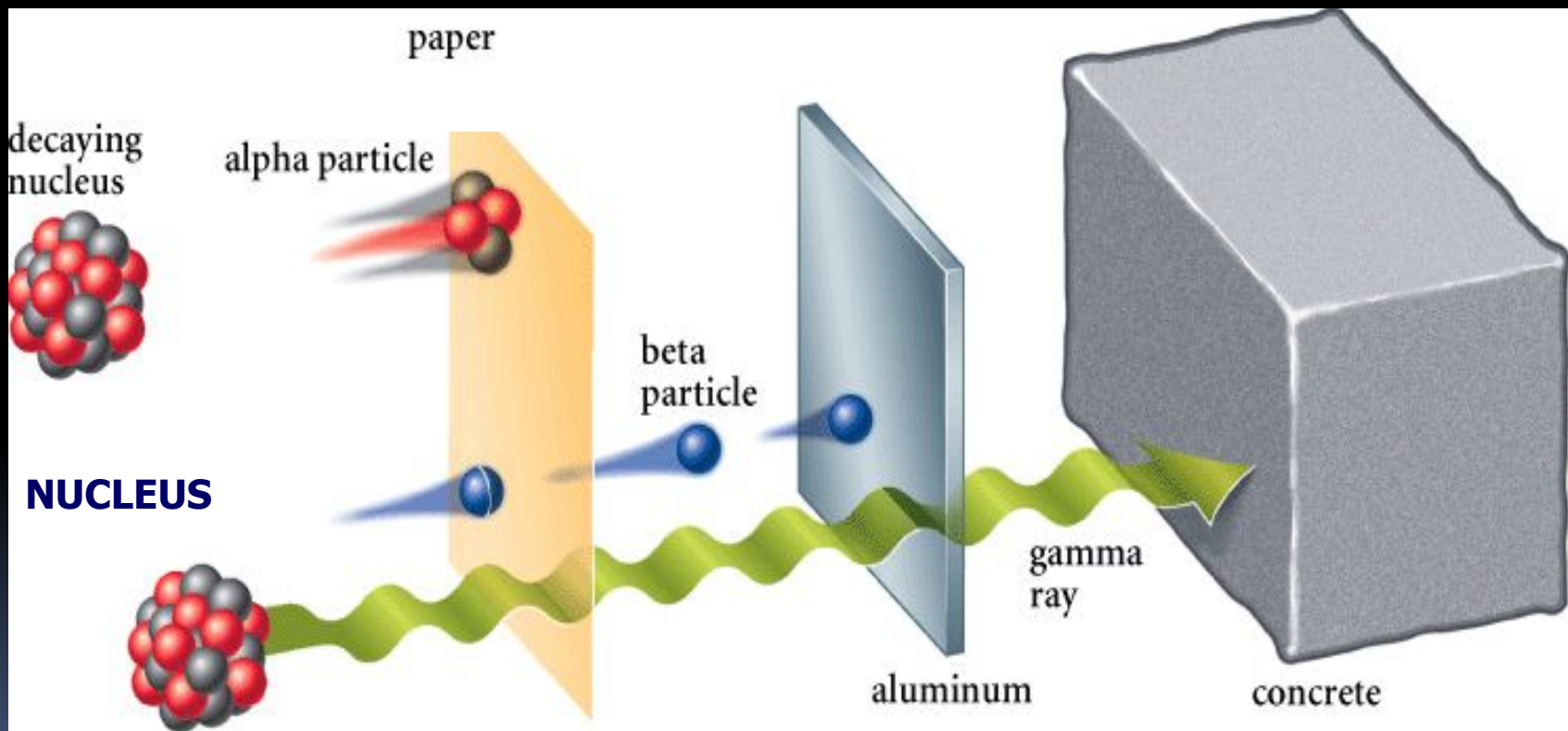
Thinking deeper: The forces in the atom

- Electrons are bound to nucleus by the Coulomb (electromagnetic) force
- Protons in nucleus are held together by the *strong* nuclear force
- Neutrons can decay into protons by weak nuclear force, emitting an electron and an anti-neutrino. The weak force is also responsible for radioactivity.

$$F = \frac{k q_1 q_2}{r^2}$$



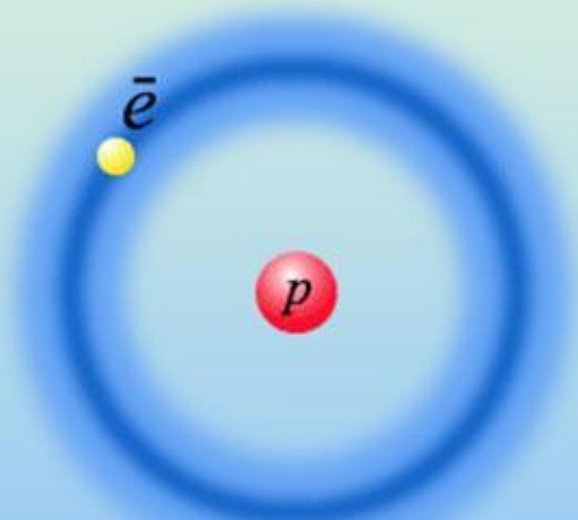
PENETRATION ABILITY OF DIFFERENT KINDS OF RADIATION



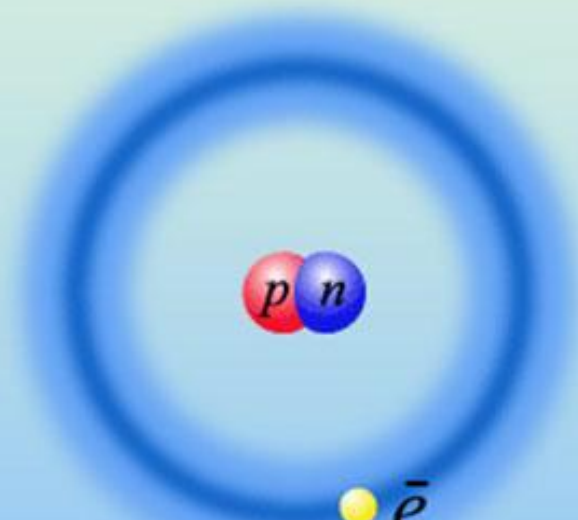
ISOTOPES

Hydrogen isotopes

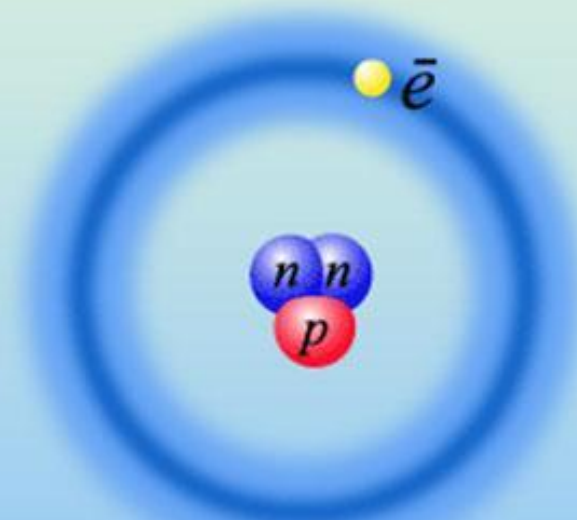
PROTIUM



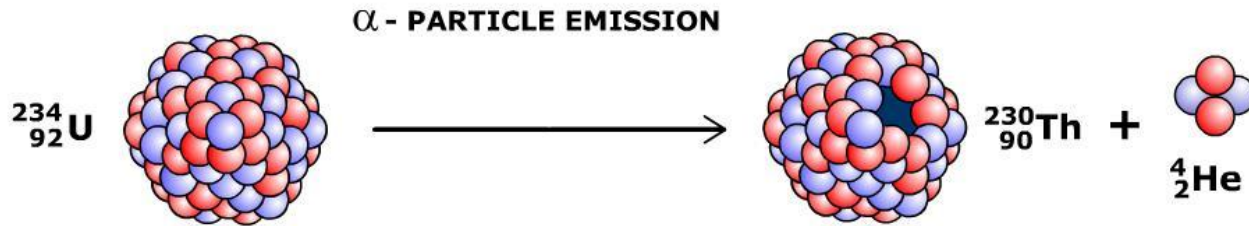
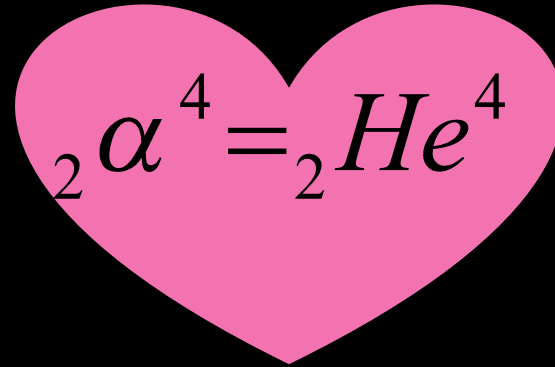
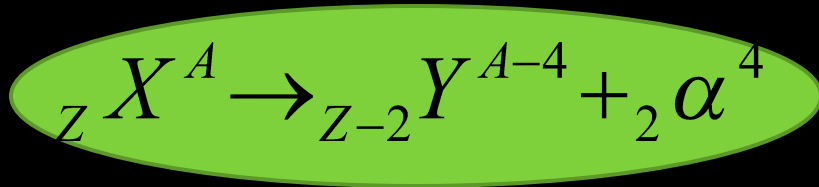
DEUTERIUM



TRITIUM



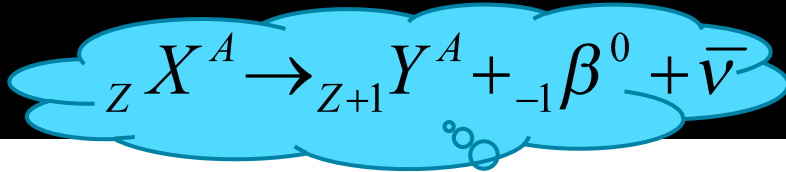
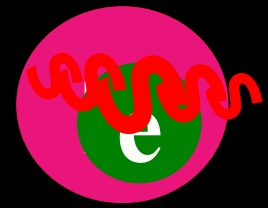
ALPHA DECAY



GAMMA-RAYS

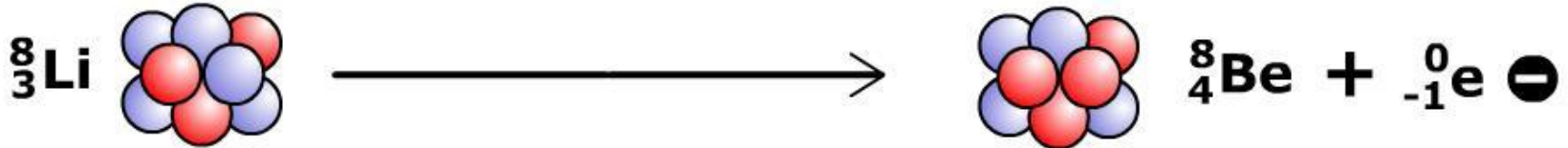


β - DECAY



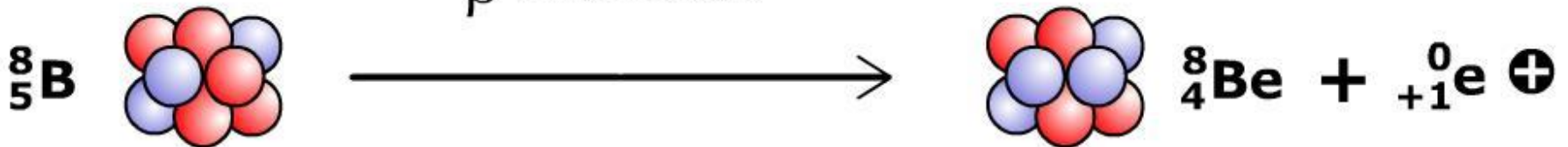
β MINUS DECAY – ELECTRON
AND ANTINEUTRINO RELEASE

β^- EMISSION

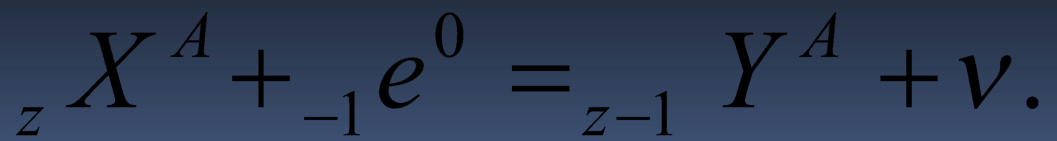


β PLUS DECAY – POSITRON AND
NEUTRINO RELEASE

β^+ EMISSION



ELECTRON CAPTURE



RADIOACTIVITY LAW

Radioactive decay law

dN – number of nuclides in radioactive decay per infinitely small period of time dt .

$\frac{dN}{dt}$ - decay rate

Radionuclide initial quantity N_0

Radioactive decay constant λ , it characterizes radionuclide of this kind decay probability

$$\frac{dN}{dt} = -\lambda \cdot N_0 \rightarrow N = N_0 \cdot e^{-\lambda \cdot t}$$

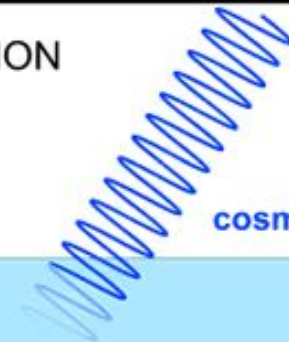
If in initial period of time ($t = 0$) there are N_0 number of nuclides after then the time t the number of nuclides left is N .

EXTERNAL AND INTERNAL IRRADIATION

NATURAL SOURCES OF RADIATION



solar radiation



cosmic radiation

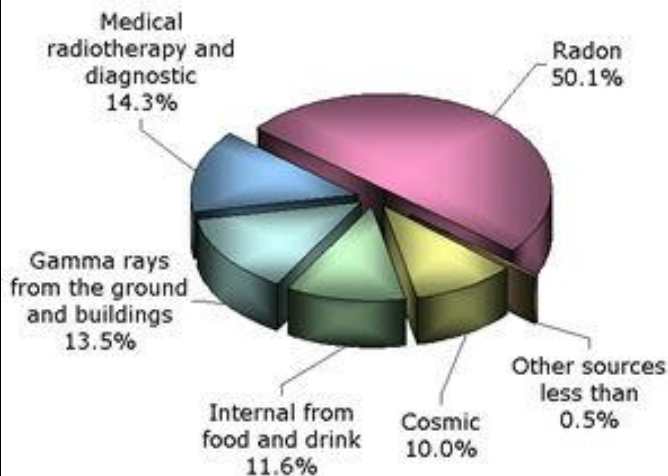
^3H ^{14}C ^7Be cosmogenic radionuclides

^{40}K internal radionuclides

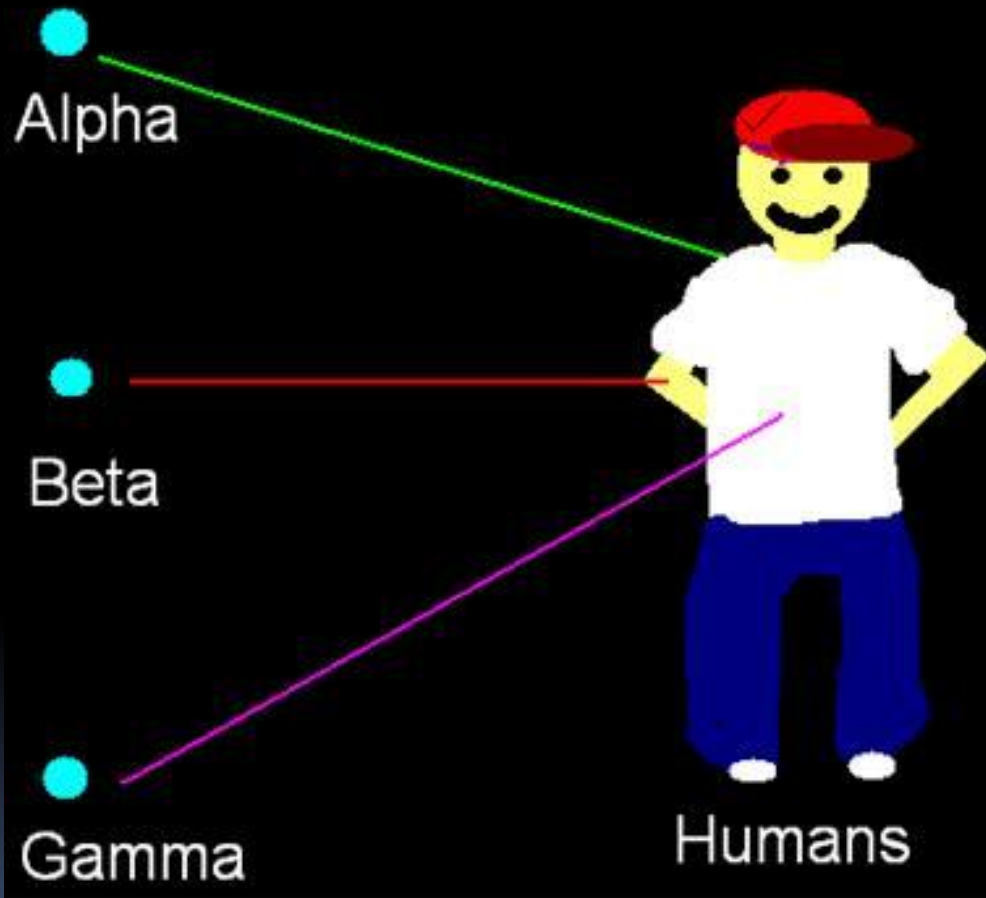
^{222}Rn inhaled radionuclides

^{226}Rn ^{232}Th ^{238}U ^{235}U terrestrial radionuclides

Be - Beryllium
C - Carbon
H - Hydrogen
K - Potassium
Rn - Radon
Th - Thorium
U - Uranium

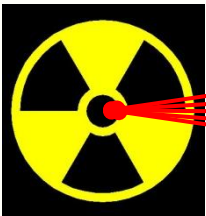


BIOLOGICAL ACTION OF RADIATION

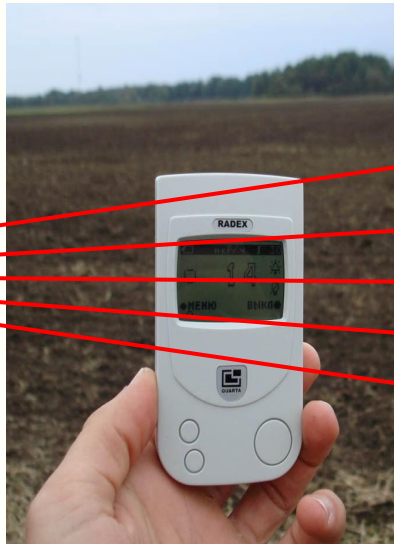


DOSIMETRY

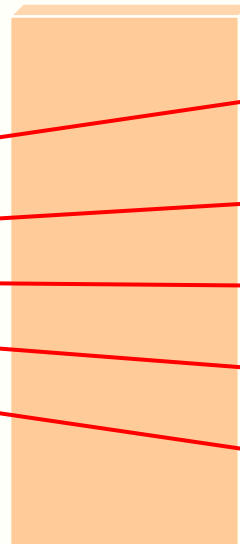
Source of radiation



Exposure dose



absorbed radiation dose



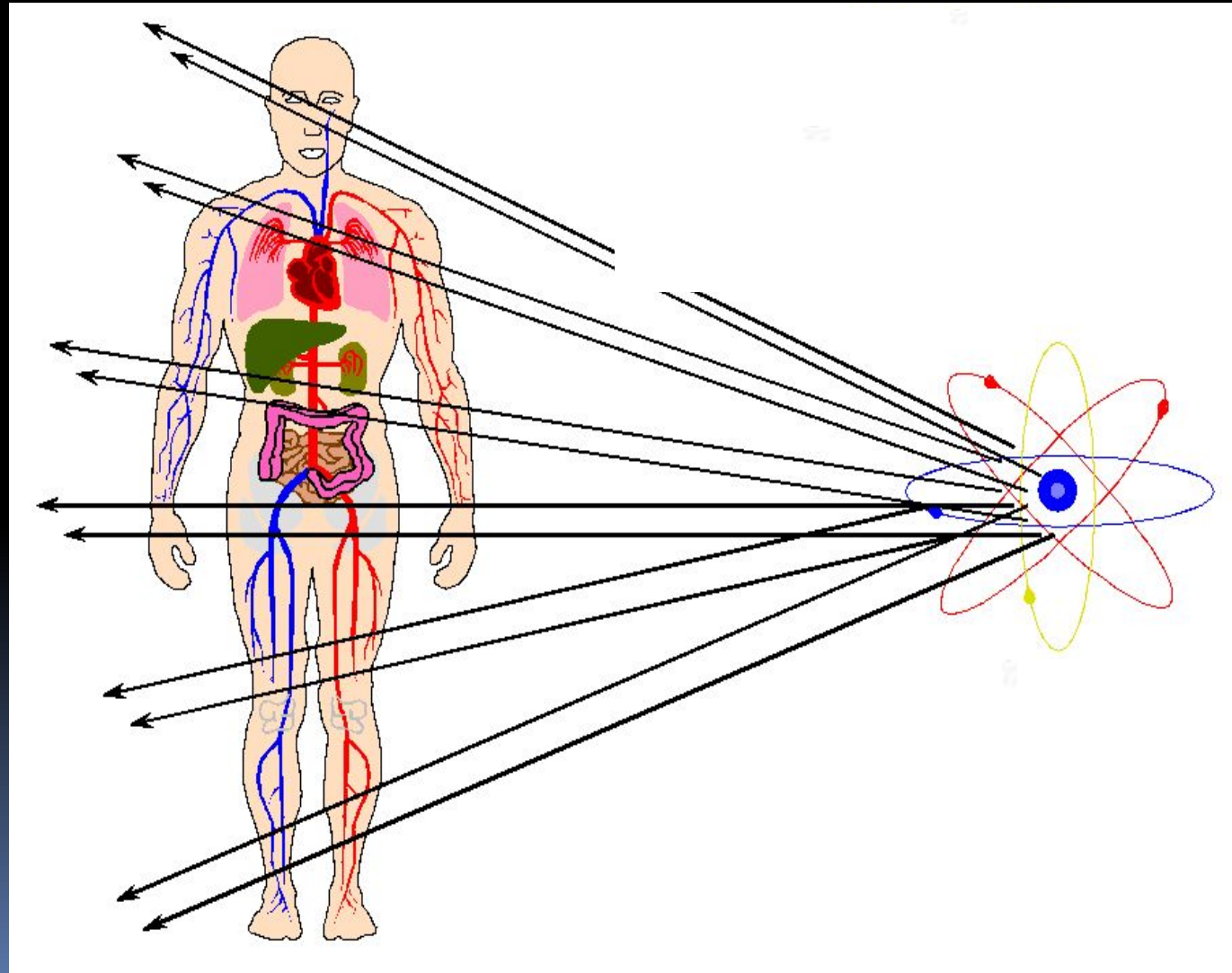
Equivalent dose



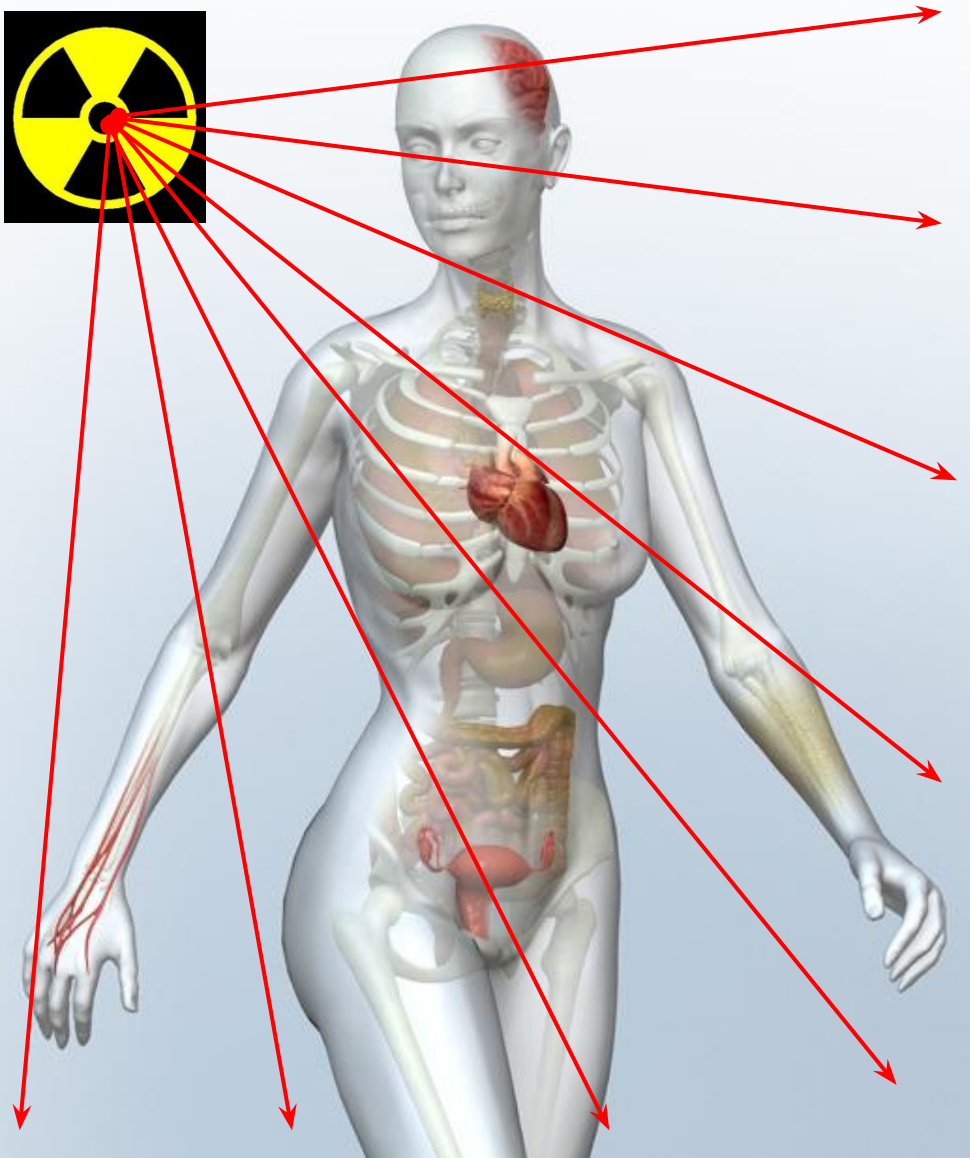
RADIATION UNITS

<p>Becquerel, Bq Curie, Cu</p>	<p>1 Bq = 1decay in sec 1 Cu = 3,7 x 10¹⁰Bq</p>	<p>Radionuclide activity – decays quantity per unit of time</p>
<p>Cl/kg (coulomb per kilogram) Rontgen, R</p>	<p>1 Cl/kg = 6,24 x 10²⁴ ion pairs/kg 1 R = 2,08 x 10¹⁵ ion pairs /m³</p>	<p>Exposure dose – quantity of charges, produced in unit of volume or mass of air due to its ionization radiation</p>
<p>Grey, Gy Rad, rad</p>	<p>1 Gy = 1 J/kg 1 rad = 0.01 Gy</p>	<p>Absorbed radiation dose – quantity of ionizing radiation energy, absorbed with unit of mass of some physical body, i.e. organism tissues</p>
<p>Sievert, Sv REM, roentgen equivalent man</p>	<p>1 Sv = 1Gy = 1J/kg (for β and γ radiation) 1 rem = 0.01 Sv = 10mSv</p>	<p>Equivalent dose – absorbed dose multiplied on coefficient of different radiation danger rate</p>

EQUIVALENT DOSE



EFFECTIVE EQUIVALENT DOSE



EED – equivalent dose calculated with different tissues radiation sensitivity taken into account. The whole organism dose equal: 1. Separate organs have their own values:

Red bone marrow – 0,12

Bones – 0,03

Mammary gland – 0,15

thyroid gland – 0,05

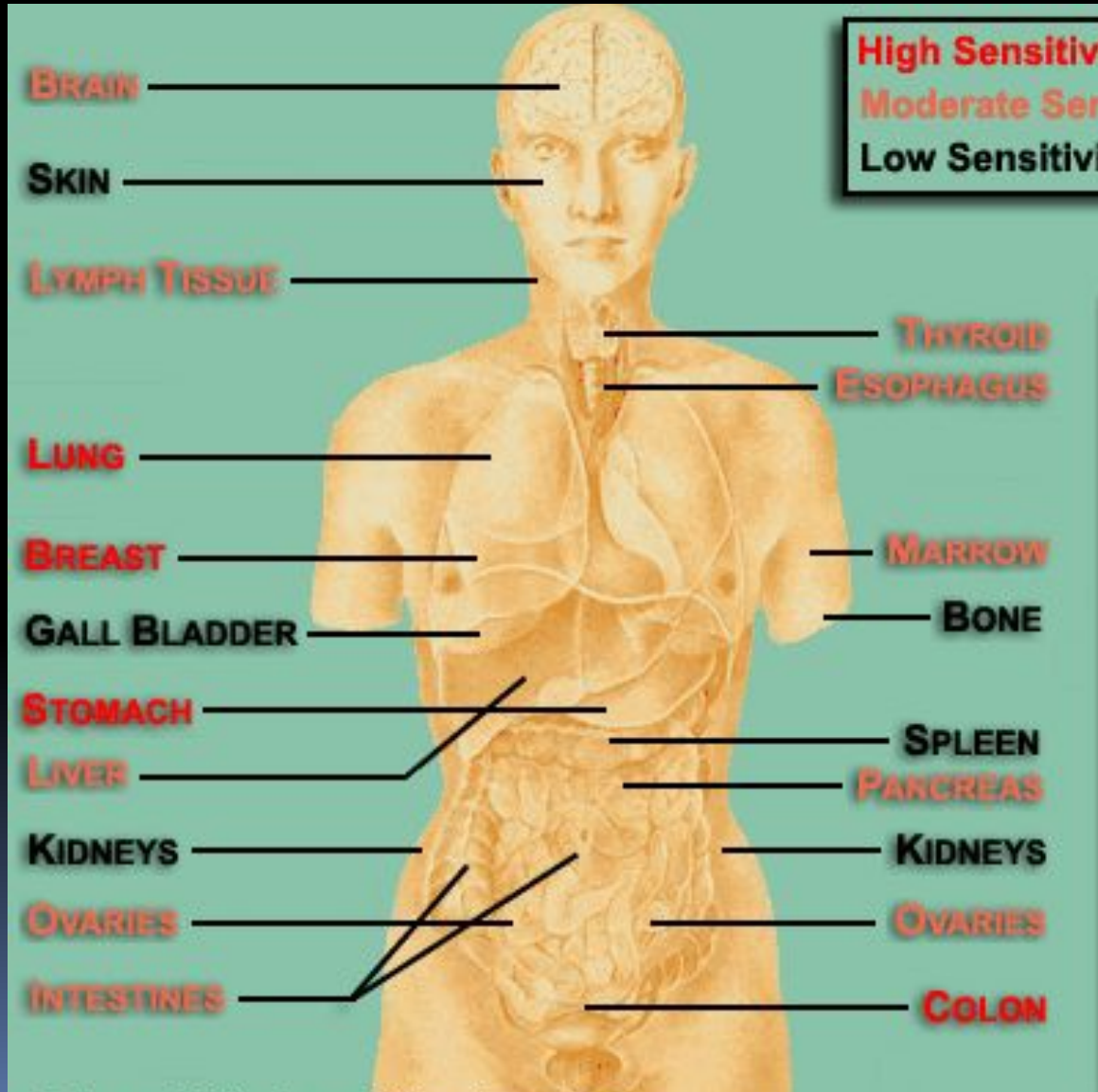
Lungs – 0,12

Stomach – 0,12

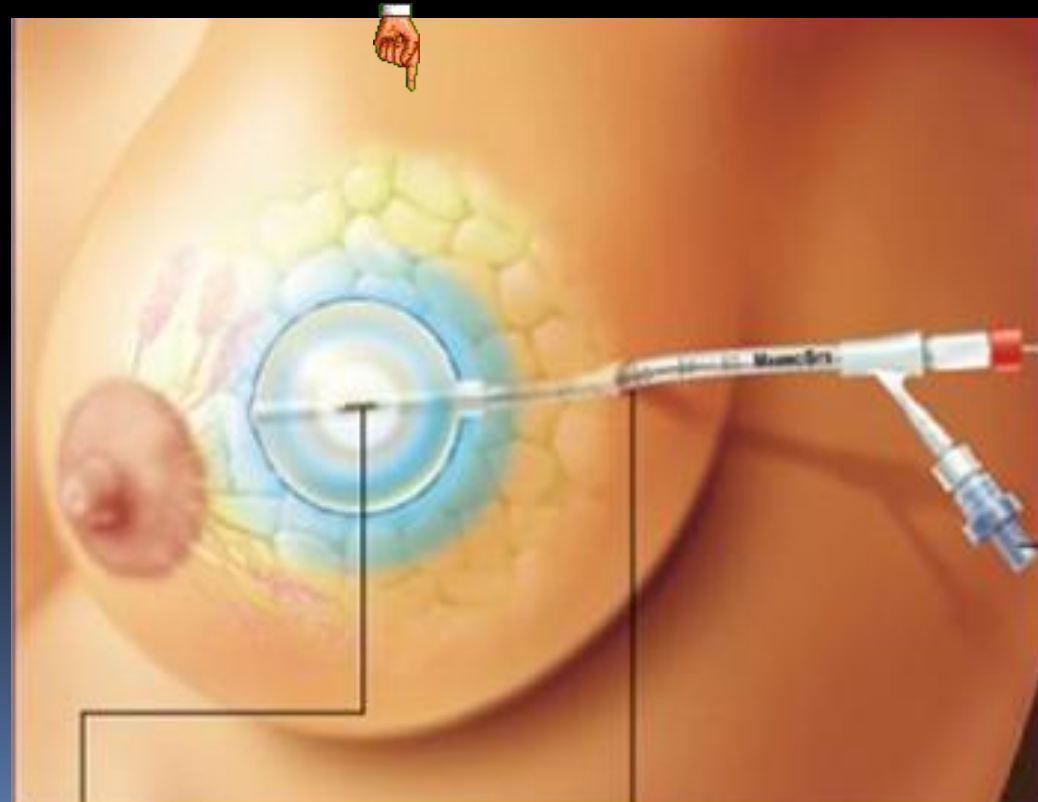
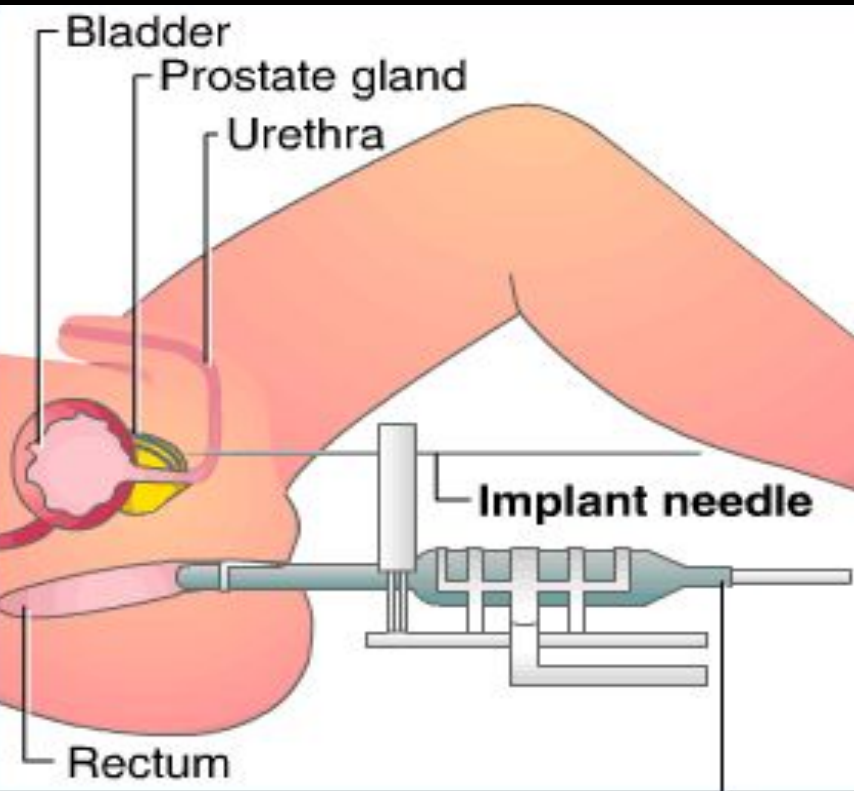
Intestine – 0,12

genital gland – 0,01

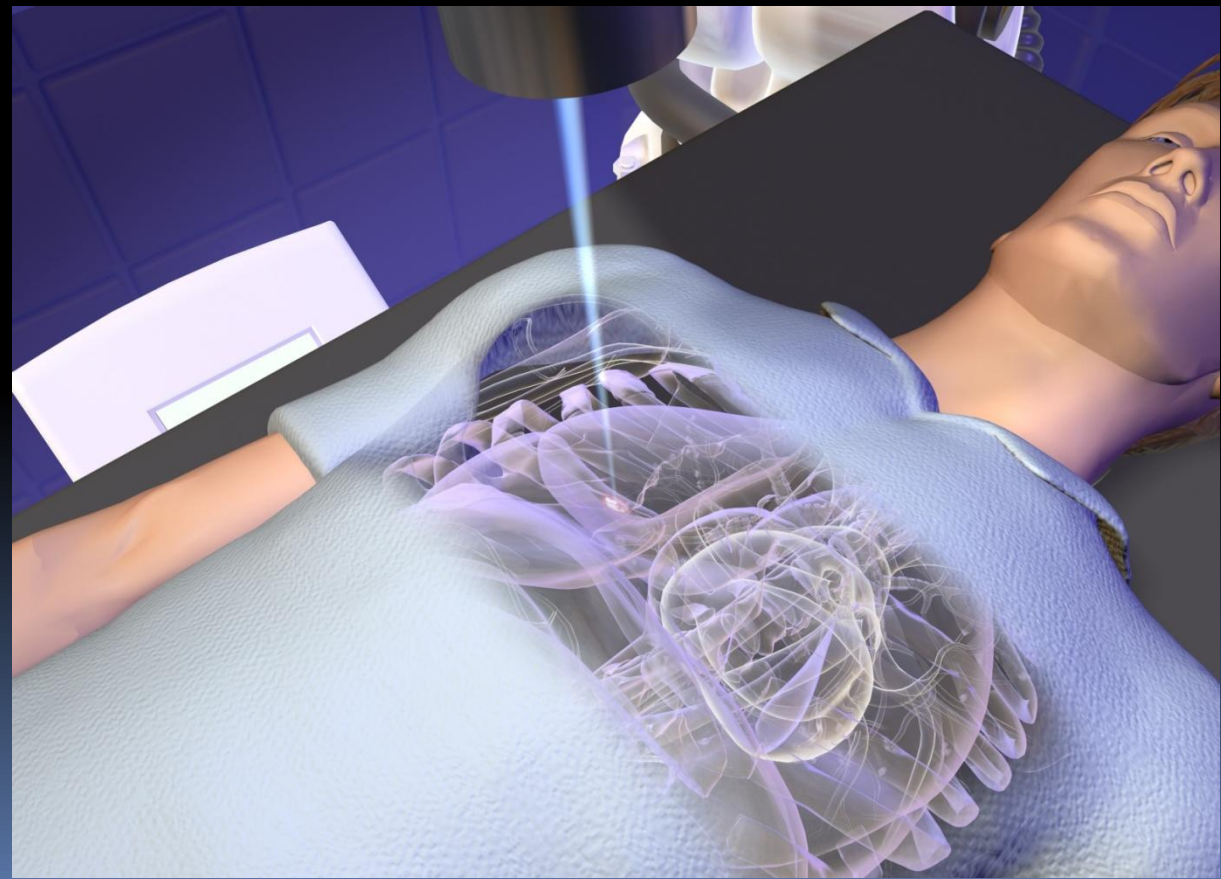
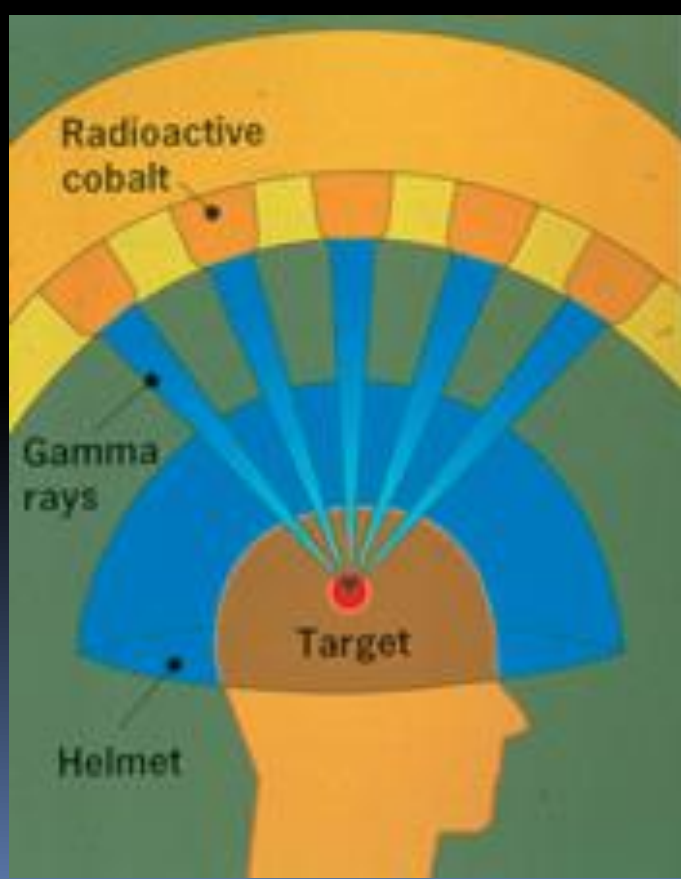
Different organs radiation sensitivity



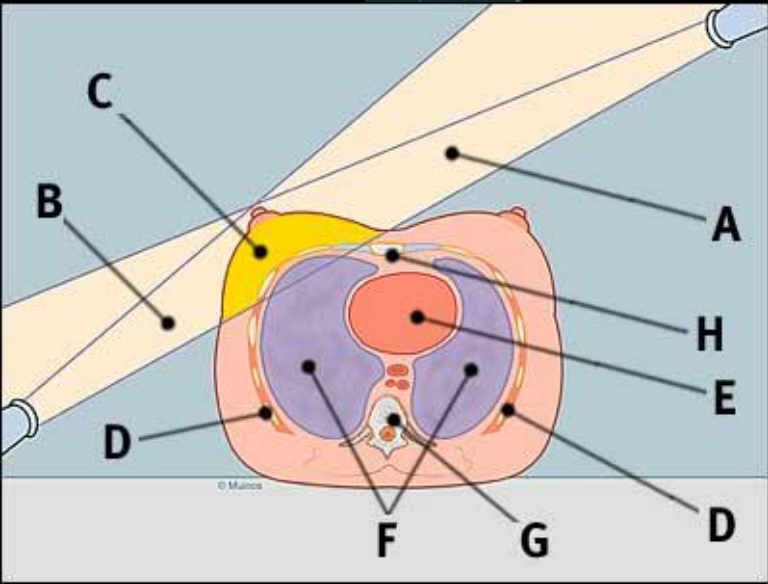
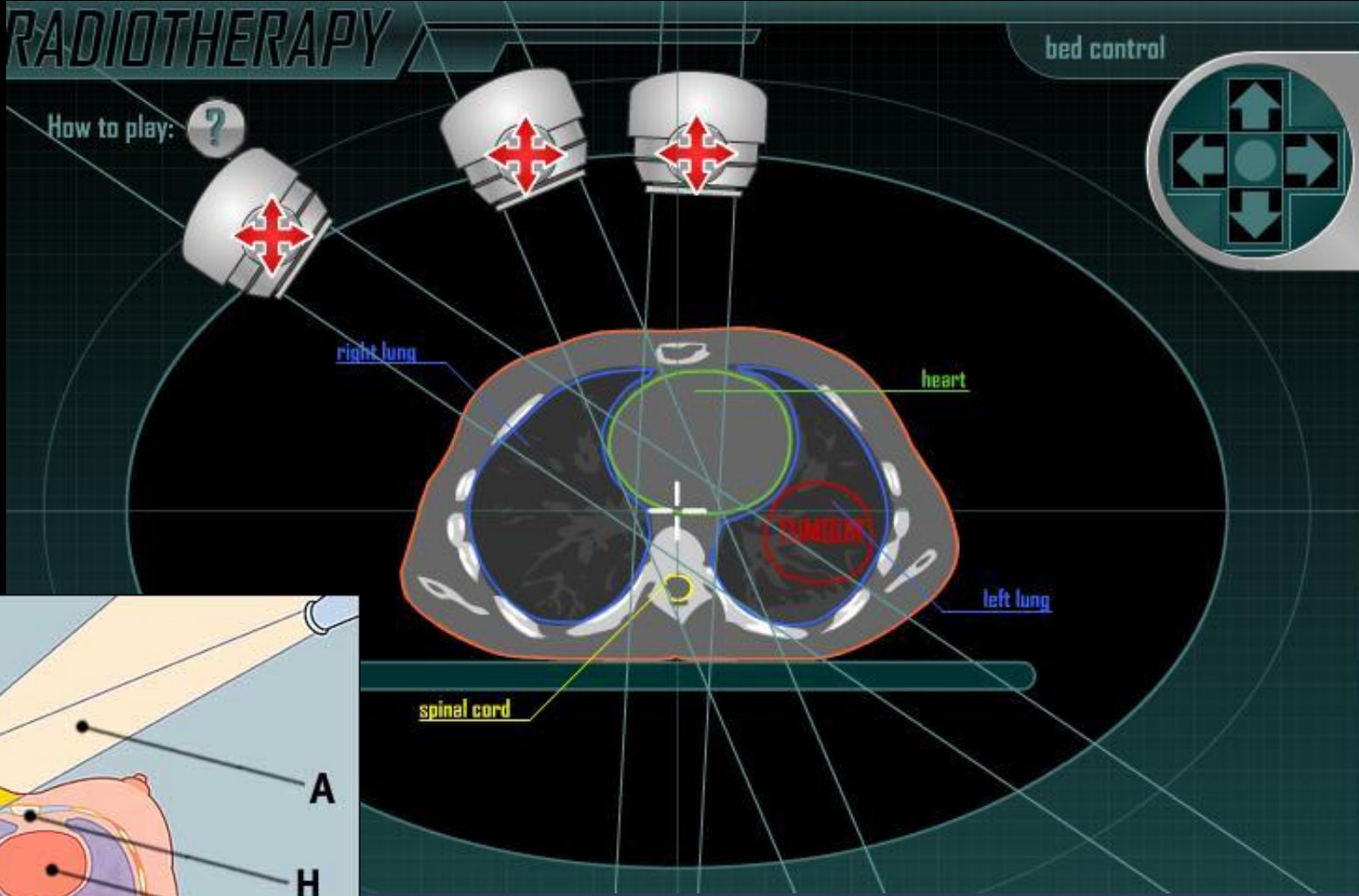
INTERNAL RADIOTHERAPY



EXTERNAL RADIOTHERAPY



EXTERNAL RADIOTHERAPY



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Cardio stimulator with plutonium battery have been working already for 34 years and had only one slight correction. The doctor described case history in New England Journal of Medicine, nuclear cardiostimulator will live longer, then its owner.