



Ask the right question to find the best answer

Is he popular?

Is he scientist?

Is there any relation of this man with today's lesson?



Can i find the information about him in the internet?

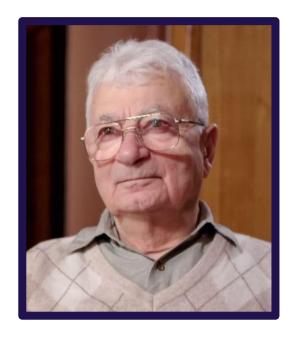
Is he alive?

Have I ever heard his name but have not seen him?





And answer is ...

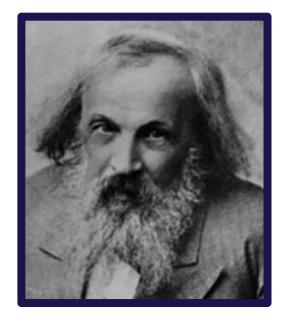


And Yuri Tsolakovich Oganessian is a Russian nuclear physicist.
The 118th chemical element oganesson was named after him in 2016, the second instance that an element was named after a living person.





GUESS THE TOPIC



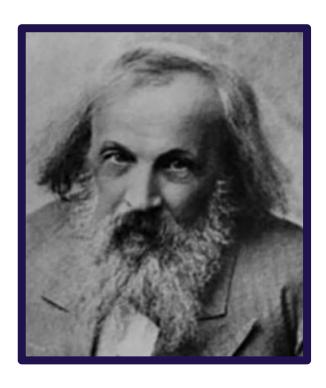
118







GUESS THE TOPIC



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опытъ системы элементовъ.
основанной на ихъ атомномъ въсъ и химическомъ сходствъ.
                  Ti=50 Zr=90 ?=180.
                   V=51 Nb= 94 Ta=182.
                  Cr=52 Mo= 96 W=186.
                  Mn=55 Rh=104,4 Pt=197,t.
                  Fe=56 Rn=104,4 Ir=198.
               NI-Co=59 PI-106,8 0-=199.
                  Cu = 63,4 Ag = 108 Hg = 200.
     Be = 9, Mg = 24 Zn = 65,2 Cd = 112
     B=11 Al=27,1 ?=68 Ur=116 An=197?
     C=12 Si-28 ?=70 Sn=118
     N=14 P=31 As=75 Sb=122 BI=210?
     0=16 S=32 Se=79,4 Te=128?
     F=19 Cl=35,6Br=80 1-127
Li=7 Na=23 K=39 Rb=85,4 Cs=133 T1=204.
           Ca=40 Sr=87, Ba=137 Pb=207.
            ?=45 Ce=92
           ?Er=56 La=94
           ?Y1-60 Di-95
           ?ln = 75,6 Th = 118?
                          A. Mengagbent
```





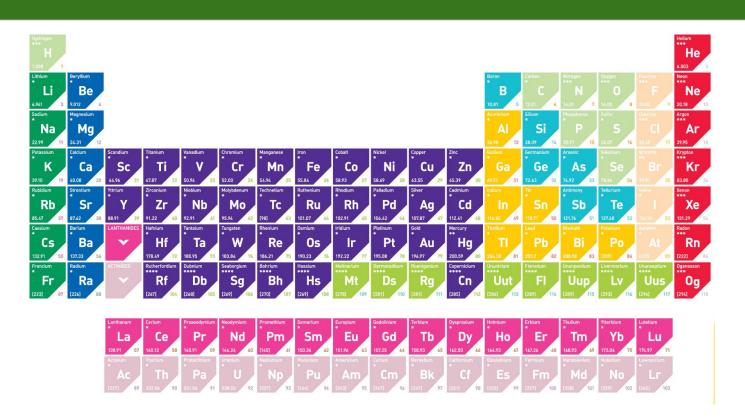
PERIODIC TABLE







MODERN PERIODIC TABLE





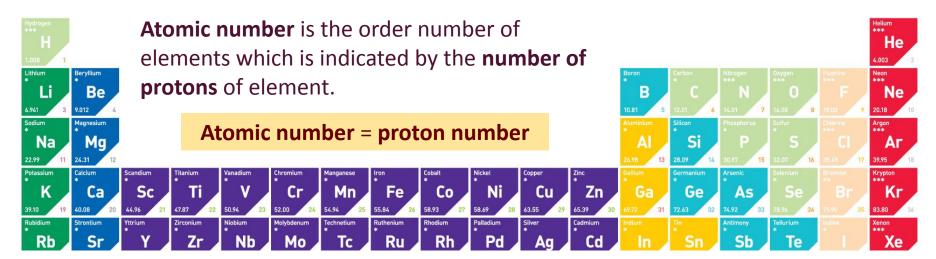


MODERN PERIODIC TABLE is very important tool for chemists and scientists.

It is easy to predict the properties of elements by using periodic table.

In PERIODIC TABLE all elements are arranged from left to right and ton to bottom in

In **PERIODIC TABLE** all elements *are arranged* from left to right and top to bottom *in order of* increasing **atomic number**

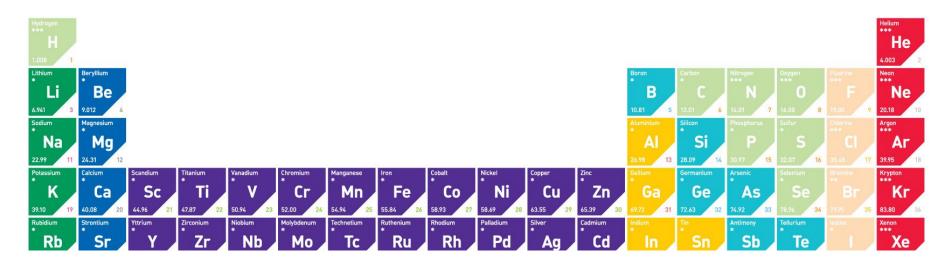






MODERN PERIODIC TABLE

- Nowadays, there are **118 chemical elements**. 94 of them are occurred **naturally**, where the remaining 24 elements are **synthetic** elements
- Simple periodic table contains **symbol**, **atomic number** and **relative atomic weight** of chemical elements

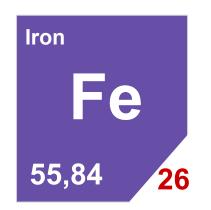






How to read the MODERN PERIODIC TABLE

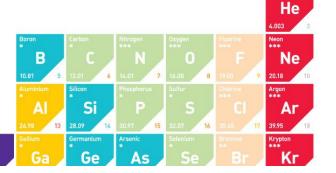






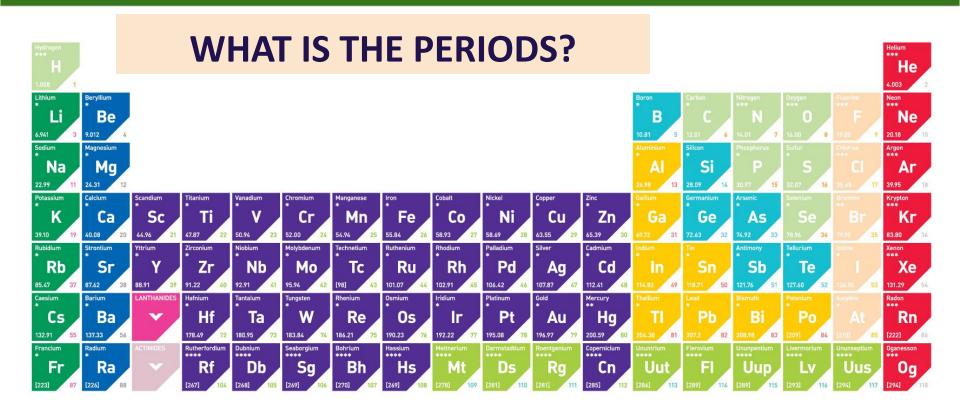


Simple periodic table contains **symbol**, **atomic number** and **relative atomic weight** of chemical elements



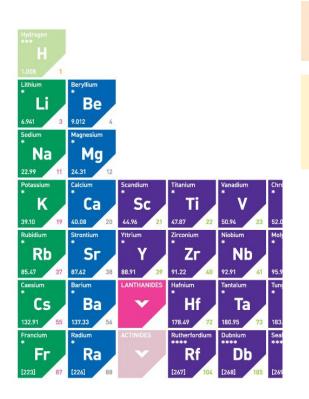






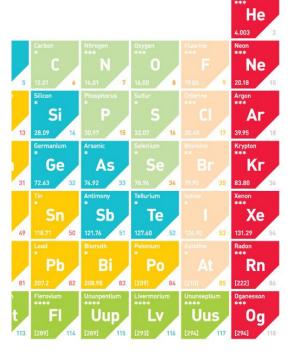






PERIODS

Horizontal rows are called **periods**



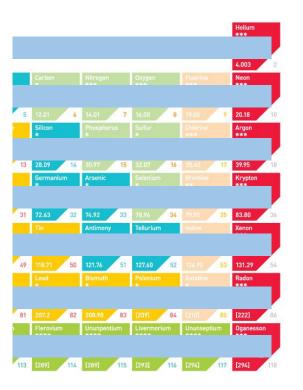






PERIODS

Horizontal rows are called **periods**









PERIODS

Horizontal rows are called **periods**

There are **seven** periods









PERIODS

Horizontal rows are called **periods**

There are **seven** periods









PERIODS

Horizontal rows are called **periods**

There are **seven** periods

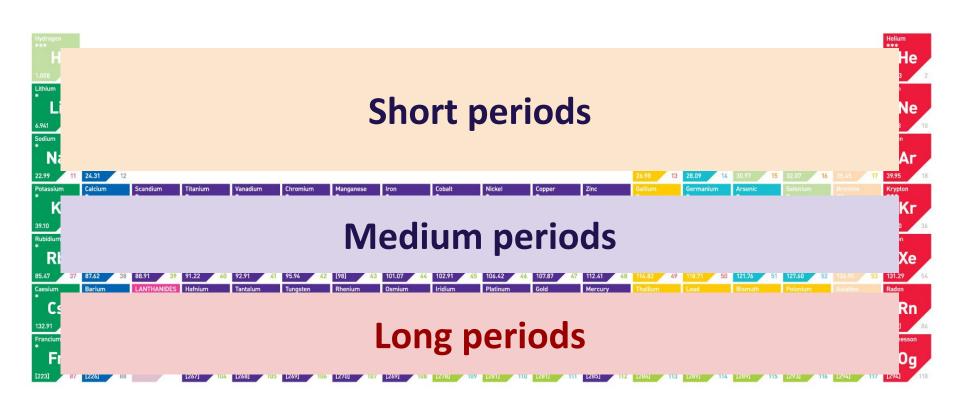
There are three short, two medium and two long periods







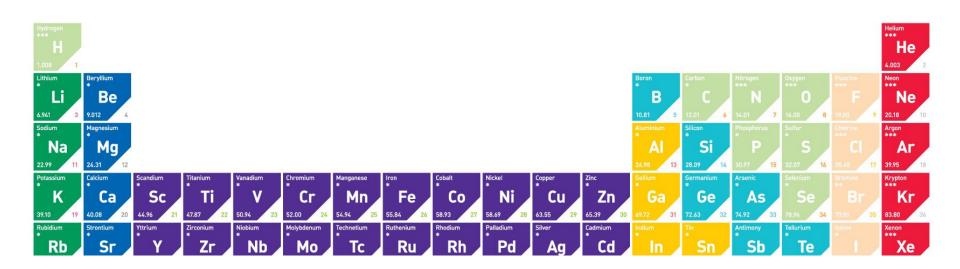
MODERN PERIODIC TABLE







WHAT IS THE GROUPS?

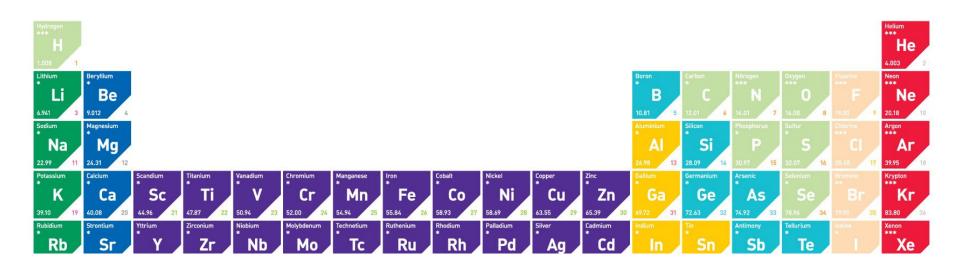






GROUPS

Vertical columns are called GROUPS

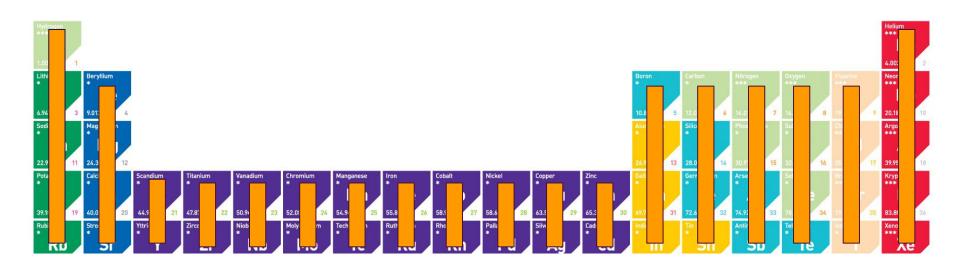






GROUPS

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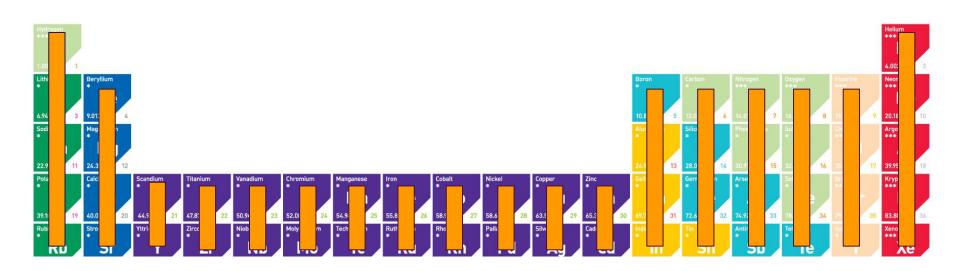




GROUPS

Vertical columns are called GROUPS

There are 18 columns



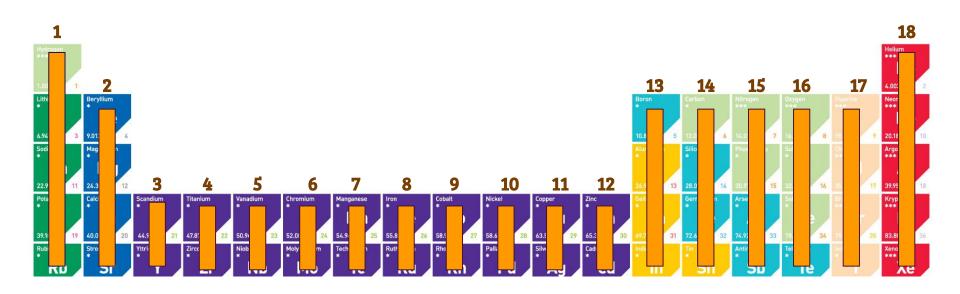




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Vertical columns are called GROUPS

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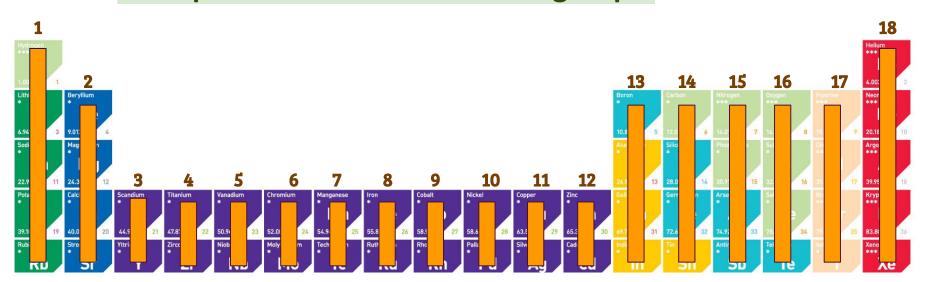


GROUPS

Vertical columns are called GROUPS

There are 18 columns

Groups are classified as A and B groups







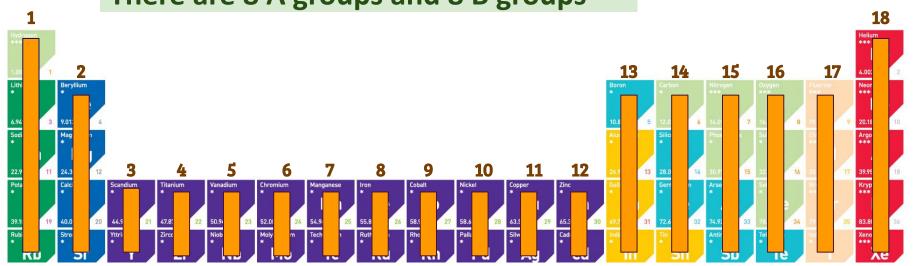
GROUPS

Vertical columns are called GROUPS

There are 18 columns

Groups are classified as A and B groups

There are 8 A groups and 8 B groups





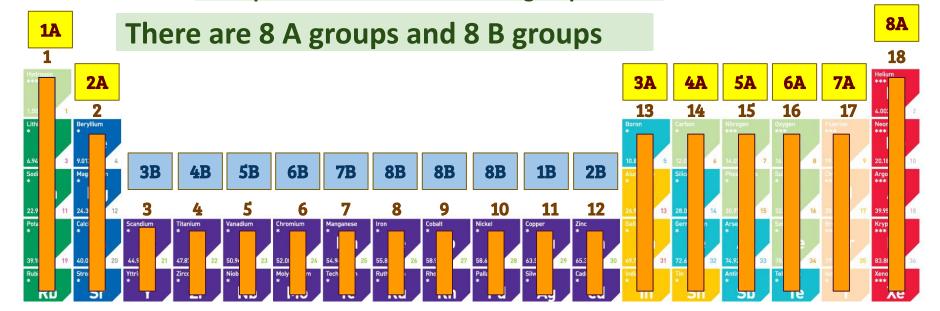


GROUPS

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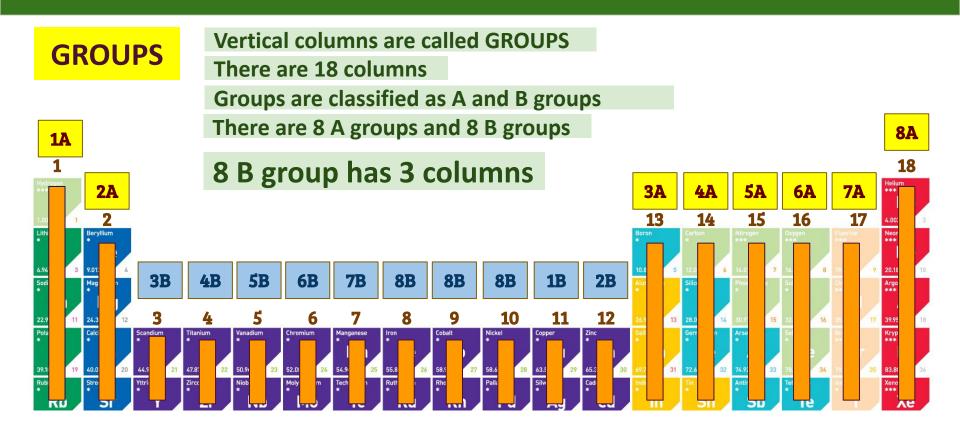
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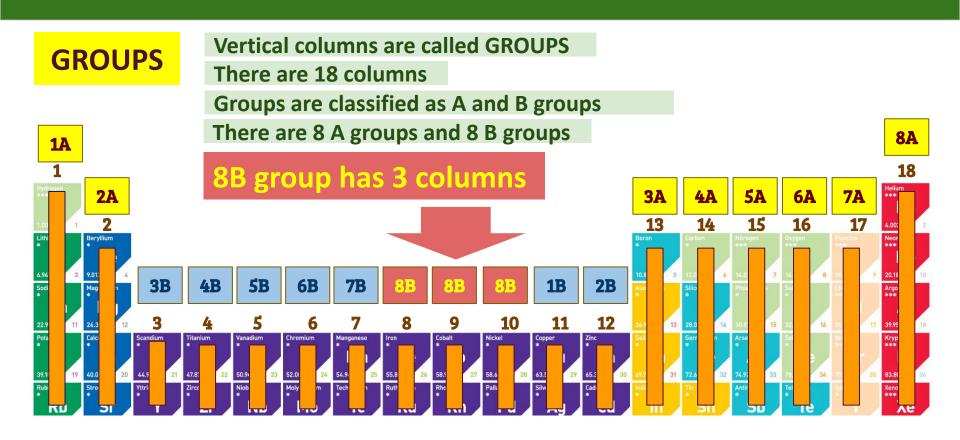












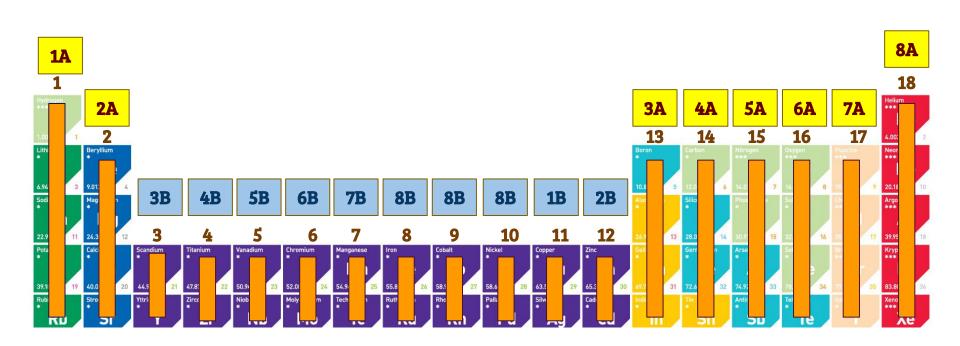




GROUPS

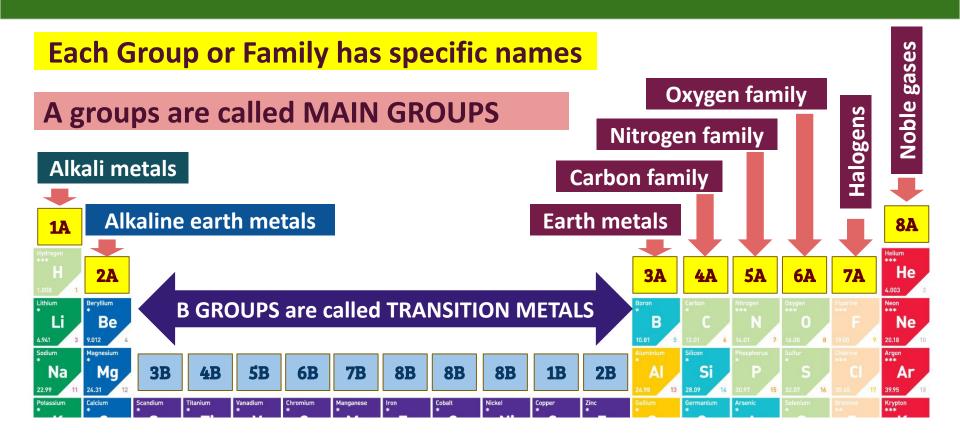
sometimes called as

FAMILY













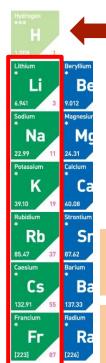
ALKALI METALS

1A group of periodic table is called Alkali metals.

This group **contains** H (hydrogen), Li (Lithium), Na (sodium), K (potassium), Rb (Rubidium), Cs (Cesium), Fr (Francium)

Alkali metals **are most active** metals. They can react even with air and water, **so** they **are stored** under the kerosene

Alkali metals **are very soft** metals, so they can be cut by knife



But hydrogen is nonmetals, it is located in 1A group because of its electronic configuration

Cesium is most active metal in 1A group

Francium is radioactive metals





MODERN PERIODIC TABLE







Worksheet 2

Find the Element activity