

# Lecture 6: Digestion

# Basics of digestion

- **Treatment** for biological waste that cannot be disposed of at a landfill
  - 2006 biodegradable waste could be placed to landfills 75%
  - 2016 only 35%
  - other methods have to be developed
- **Digestion facilities in Finland**
  - Mainly at waste water plants for sludge treatment (~ 15 facilities)
  - A few facilities for municipal bio-waste treatment (Stormossen, Laihia)
  - A few industrial waste facilities
  - A few large facilities for farm waste (Close to Turku, Juva....)
  - Several facilities for farm waste treatment
  - The facilities in Finland produce over 25 mill. m<sup>3</sup> biogas
  - Biogas can be used for energy production or fuel for vehicles
- **Facility sizes** vary from private farm reactors (< 100 m<sup>3</sup>) to Helsinki Water reactor (10 000 m<sup>3</sup>)

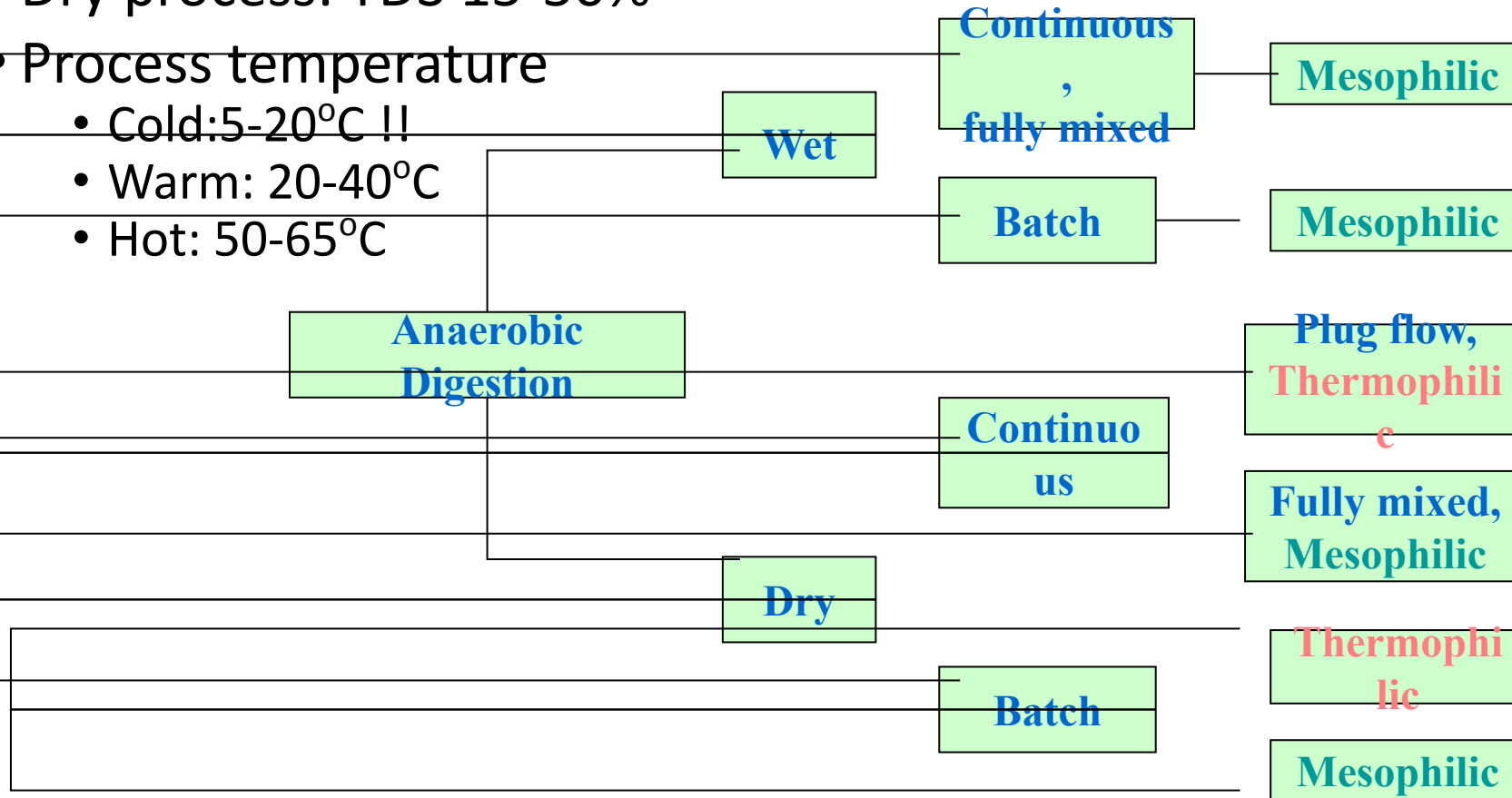
# Classification of anaerobic processes

- Wet process: total dry solids (TDS) 5-15%

- Dry process: TDS 15-50%

- Process temperature

- Cold: 5-20°C !!
- Warm: 20-40°C
- Hot: 50-65°C



# Digestion process

Biological reactions in the digestion are similar to those in anaerobic landfill

**Hydrolysis:** fermentative bacteria hydrolyze complicated organic compounds into soluble organics more available for the next stage

- Enzymes produced by hydrolytic bacteria decompose and liquefy carbohydrates, cellulose, proteins and fats
- Rate limited: decomposing the complex compounds like cellulose
- Rate governed by
  - Substrate availability
  - Bacterial population density
  - Temperature and pH

**Acidogenesis (acidogenesis and acetogenesis):** products of the hydrolysis are further processed by bacteria

- Main products: acetic, lactic and propionic acids
  - Acetic acid is produced from monomers
  - Volatile fatty acids (VFA) are produced from protein, fat and carbohydrate components
- Some gases ( $\text{CO}_2$ ,  $\text{H}_2$ ) and methanol are produced
- pH falls
- Products depend on feedstock, bacteria species and environmental conditions

# Digestion process

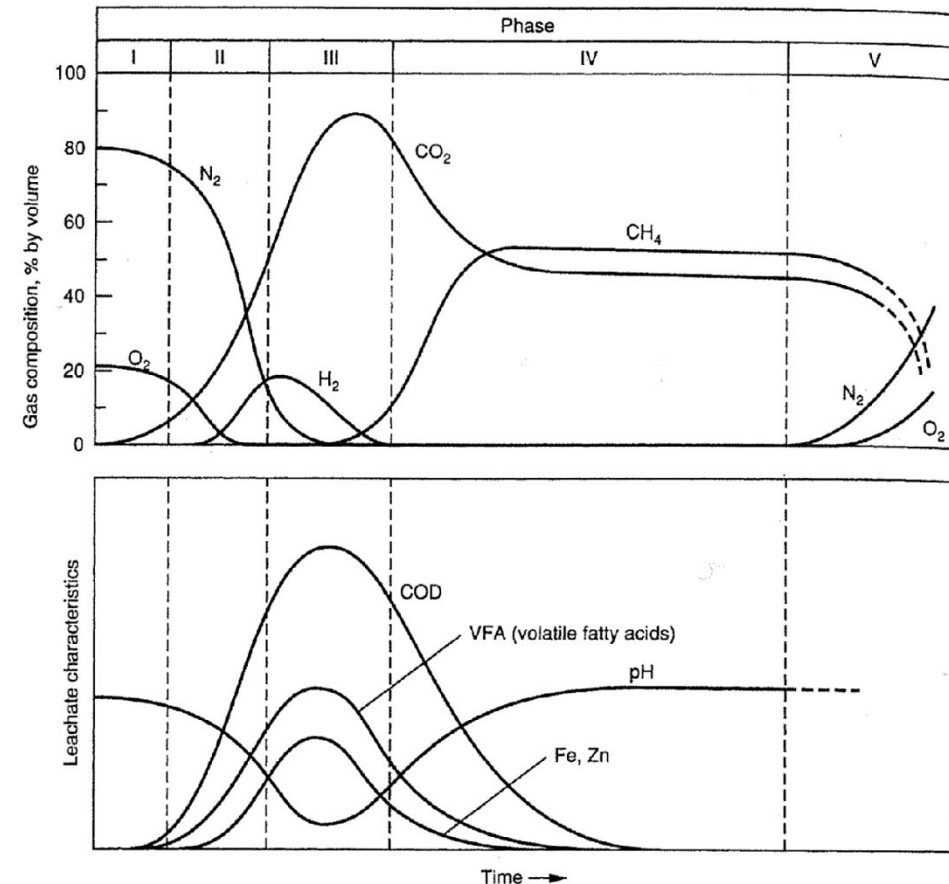
**Methanogenesis:** methane - forming bacteria produces methane from the products of previous stage (HAc, MeOH, CO<sub>2</sub>, H<sub>2</sub>)

- Acetic acid + acetate  $\square$  75% of CH<sub>4</sub>
  - CH<sub>3</sub>COOH  $\square$  CH<sub>4</sub> + CO<sub>2</sub>
- Methanol and hydrogen can be used, too
  - CH<sub>3</sub>OH + H<sub>2</sub>  $\square$  CH<sub>4</sub> + H<sub>2</sub>O
- Carbon dioxide and hydrogen produce methane, too
  - CO<sub>2</sub> + 4H<sub>2</sub>  $\square$  CH<sub>4</sub> + 2H<sub>2</sub>O
- Converting volatile fatty acids into methane maintains higher pH
  - pH stays at 6,6 – 7,0 (mild acidic)
  - Problems arise if pH <6,4
  - Volatile fatty acids would be harmful for fertilizer use of the final product

# Gas formation in anaerobic processes

See anaerobic processes in landfills for more detailed description

- **Phase I**
  - Atmospheric levels of  $N_2$  and  $O_2$
- **Phase II**
  - $N_2$  falls to 10%
  - Oxygen is depleted
  - Fatty acids and  $CO_2$  formed
- **Phase III**
  - $CO_2$  falls to 40%
  - $CH_4$  rises to 60%
- **Phase IV**
  - Plateau:  $CO_2$  40% and  $CH_4$  60%
- **Phase V**
  - $CO_2$  and  $CH_4$  production to  $\sim 0$



**FIGURE 14.6** Generalized phases in the generation of landfill gases (I—Initial Adjustment, II—Transition Phase, III—Acid Phase, IV—Methane Fermentation, and V—Maturation Phase). (Adapted from Farquhar and Rovers, 1973; Parker, 1983; Pohland, 1987; and Pohland, 1991.)

# Process variables

- Internal environment has to be optimal for fast reactions in reducing the volume of waste and producing biogas efficiently
- Physical conditions
  - Mixing
  - Temperature has to be relatively constant
    - Mesophilic (33-37 °C) more used in Finland
      - 21 days
    - Thermophilic (54 °C)
      - Faster: eg. 14 days
      - Destroys pathogens better
  - Others: Retention period, wetness, feedstock characteristic, digester loading, bacterial population
- Chemical conditions
  - pH should stay relatively high
  - Alkalinity works as a buffer against acidity
  - Volatile fatty acids concentration affects pH