

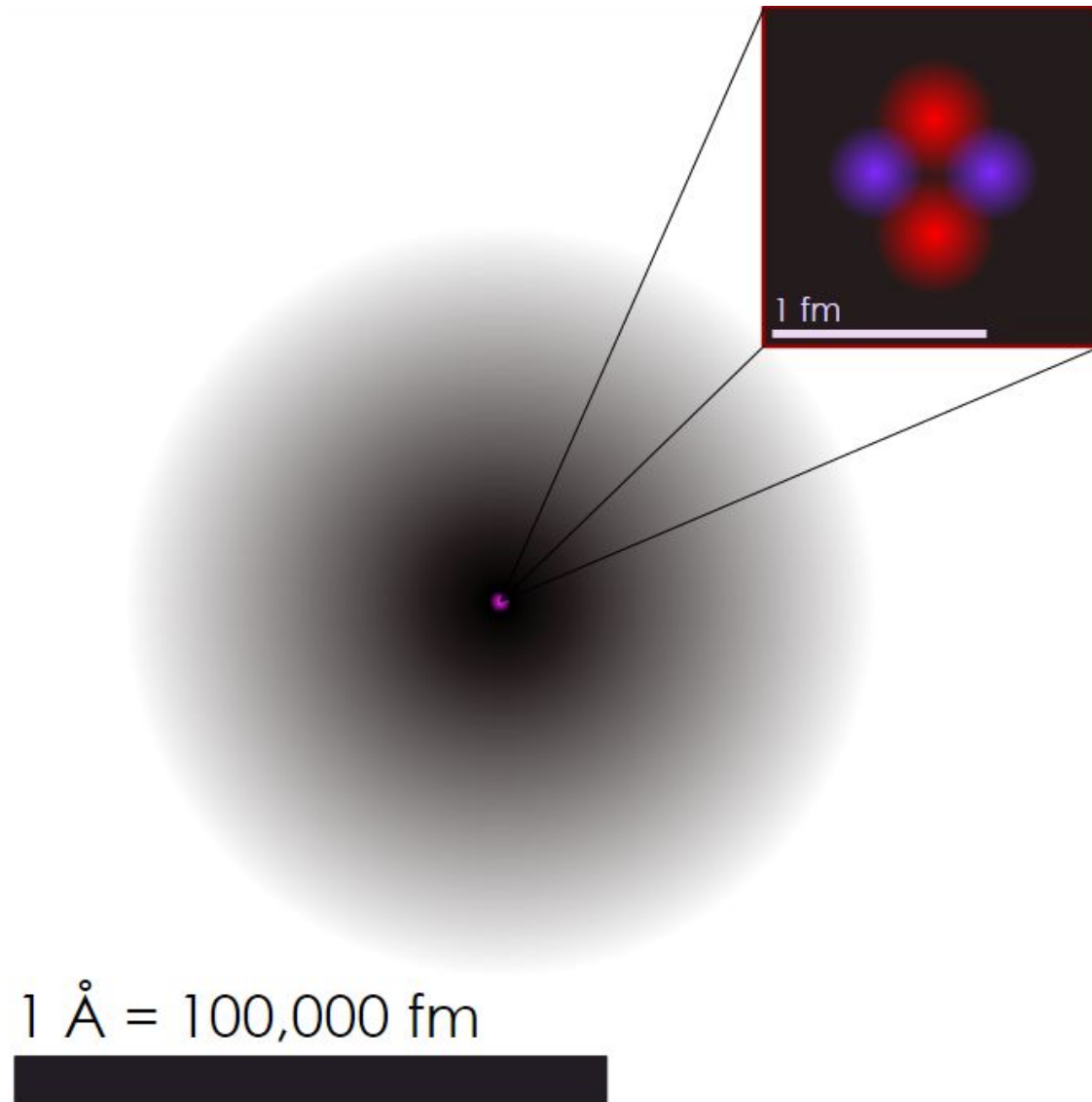
Periodic Table: It's Elementary!

Mikhail V. Kurushkin, PhD

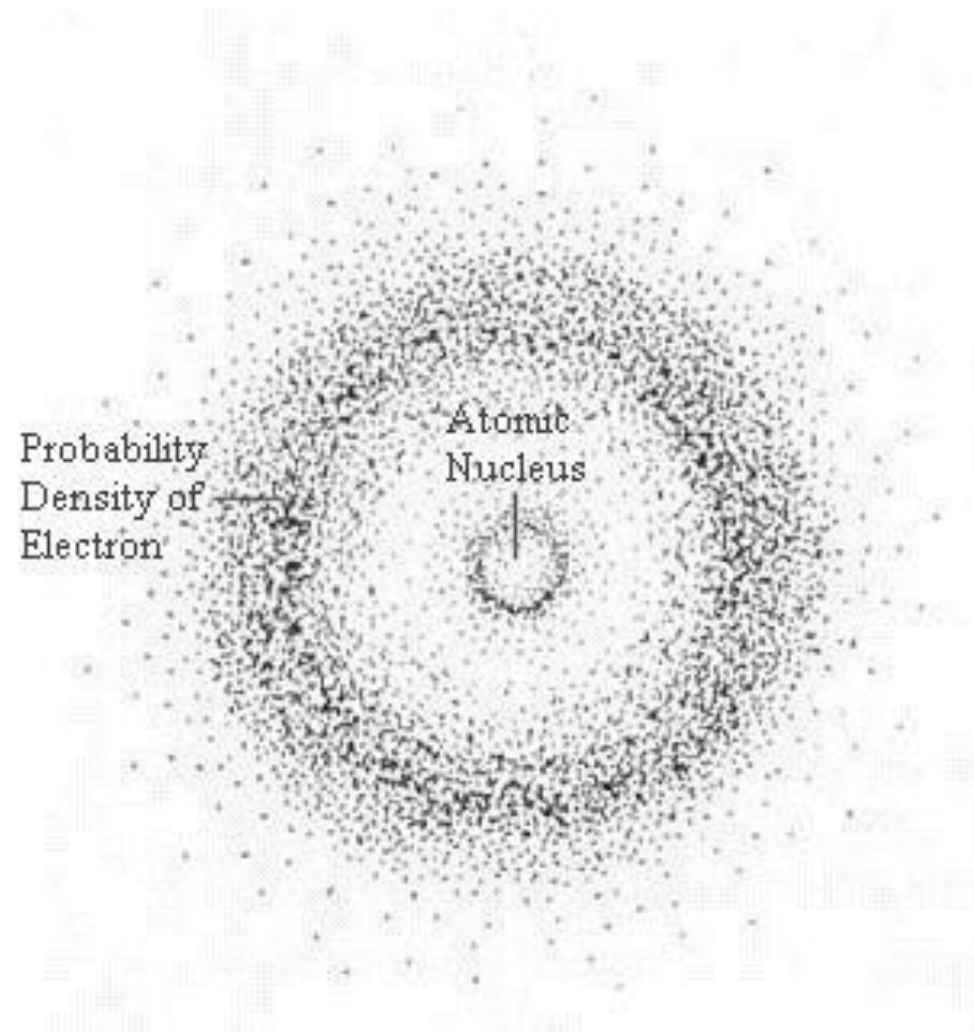


April 8-9, 2017, Sochi

Periodic Table: It's Elementary!

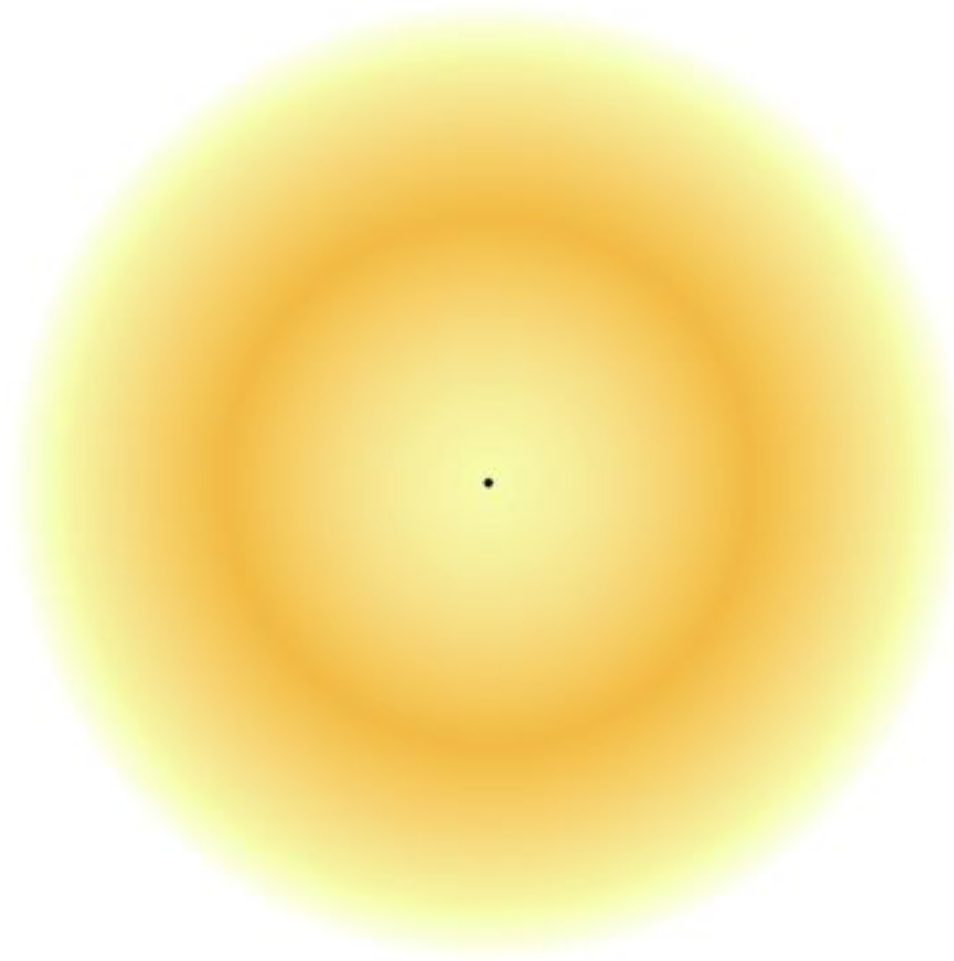


Periodic Table: It's Elementary!




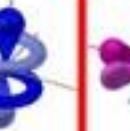


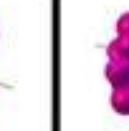
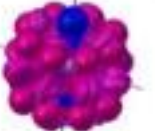



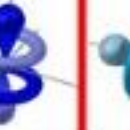


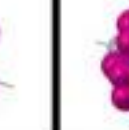
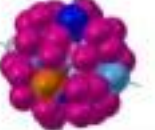


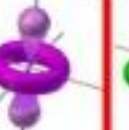





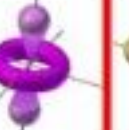











Periodic Table: It's Elementary!

What An Electron Is



Periodic Table: It's Elementary!

TYPE	SET	INDIVIDUAL						COLLECTIVE	
f	Cubic								
	General								
d	Common								
	"Tri-torus"								
p									
s									

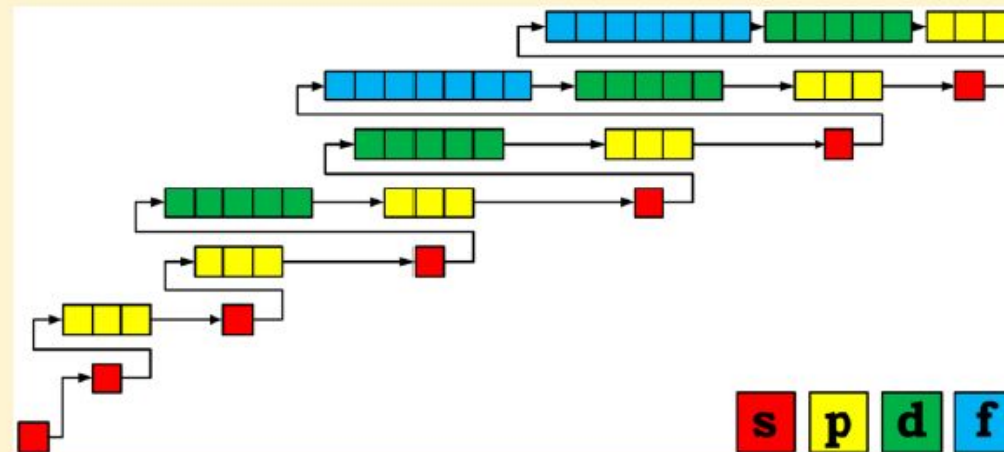
Teaching Atomic Structure: Madelung's and Hund's Rules in One Chart

Mikhail Kurushkin*

Saint Petersburg State Polytechnical University, 29 Polytechnicheskaya Street, Saint Petersburg 195251, Russian Federation

S Supporting Information

ABSTRACT: A diagram that represents electron levels and sublevels with the corresponding values of quantum numbers has been designed. The main purpose of the diagram is the explanation of the atomic structure to the students according to Hund's rule and especially Madelung's rule. The diagram represents a specific arrangement of all sublevels' orbital diagrams in order to enhance the understanding of said rules.



KEYWORDS: High School/Introductory Chemistry, Physical Chemistry, Textbooks/Reference Books, Atomic Properties/Structure

n	1	2	3	4	5	6	7
m	0	+1 0 -1	+2 +1 0 -1 -2	+3 +2 +1 0 -1 -2 -3	+4 +3 +2 +1 0 -1 -2 -3 -4	+5 +4 +3 +2 +1 0 -1 -2 -3 -4 -5	+6 +5 +4 +3 +2 +1 0 -1 -2 -3 -4 -5 -6
13							
12							7i
11							7h
10							7g
9							7f
8							7d
7							7p
6							7s
5							
4							
3							
2							
1							
$n+1$	2/2	8/8	18/18	32/32	32/50	18/72	8/98

Periodic Table: It's Elementary!

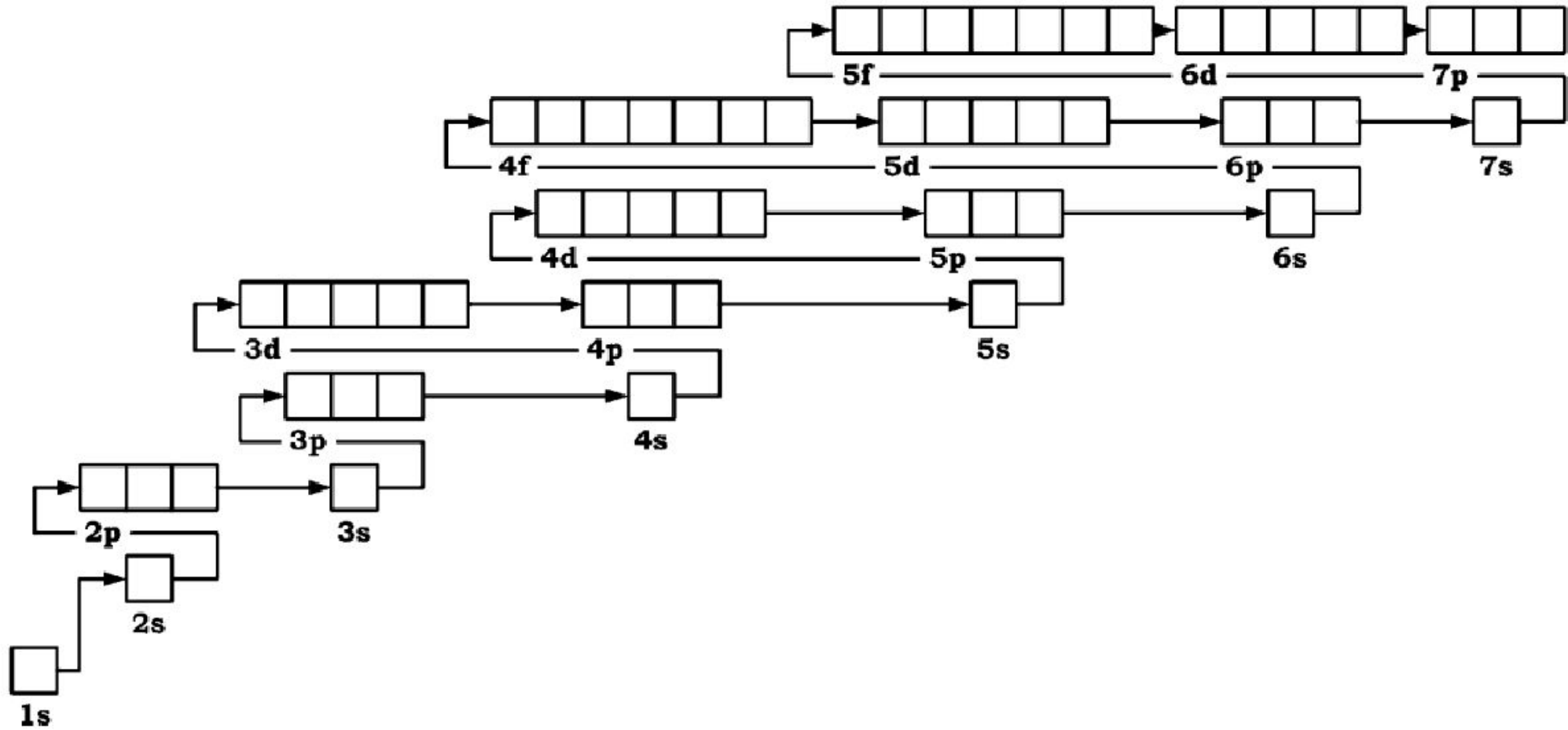
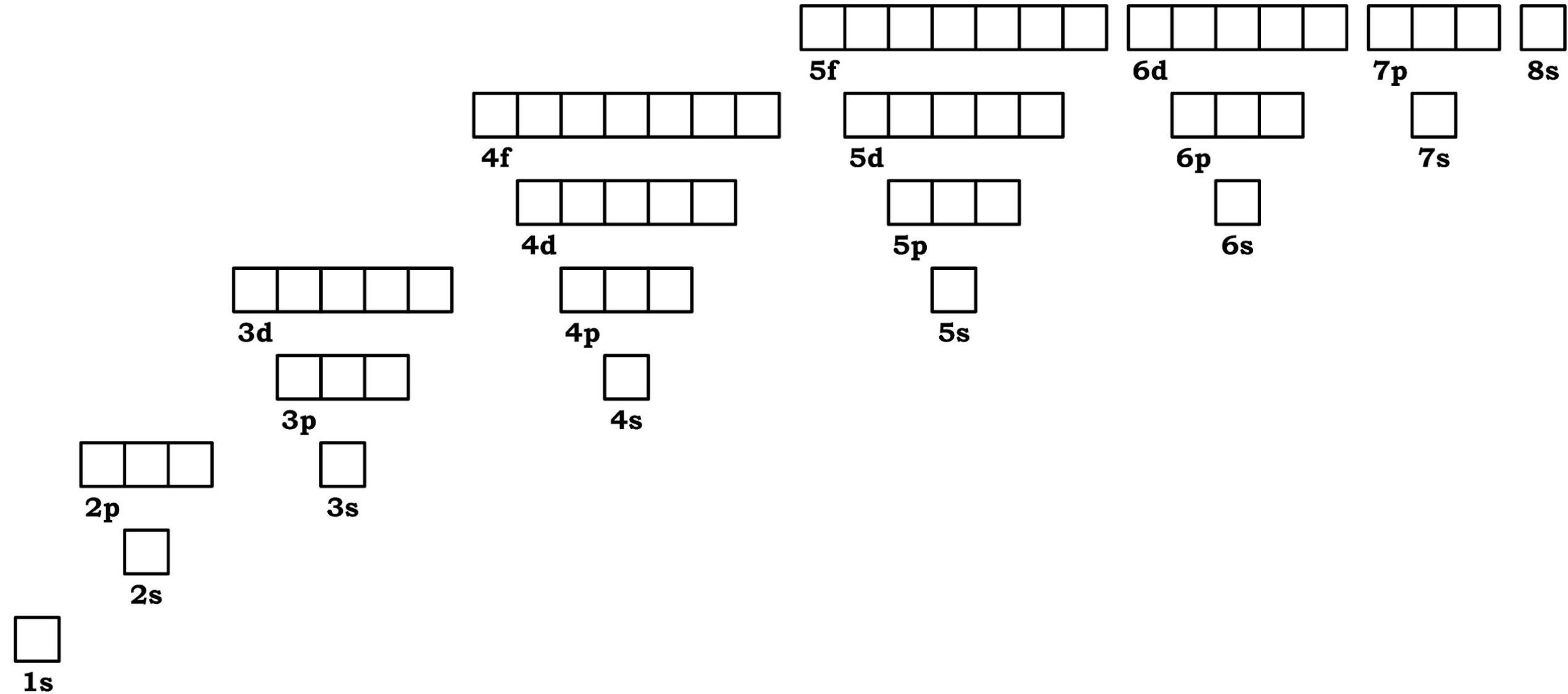
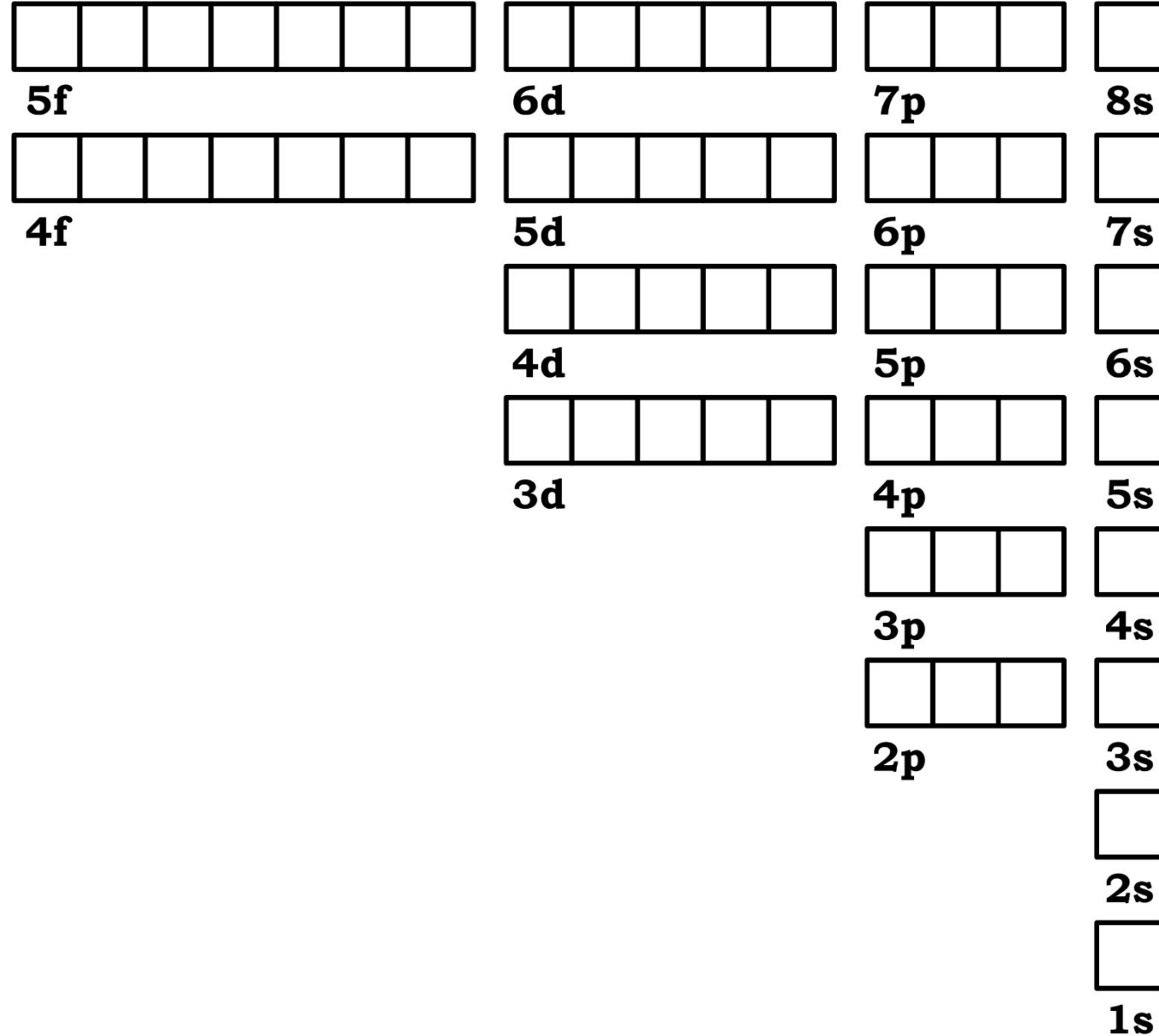


Figure 2. Order for filling the sublevels with electrons according to Madelung's rule.

Periodic Table: It's Elementary!



Periodic Table: It's Elementary!



Periodic Table: It's Elementary!

							f		d			p		s	
															8
															7
															6
															5
															4
															3
															2
															1

Periodic Table: It's Elementary!

																	1
																	2
																	3
																	4
																	5
																	6
																	7
																	8

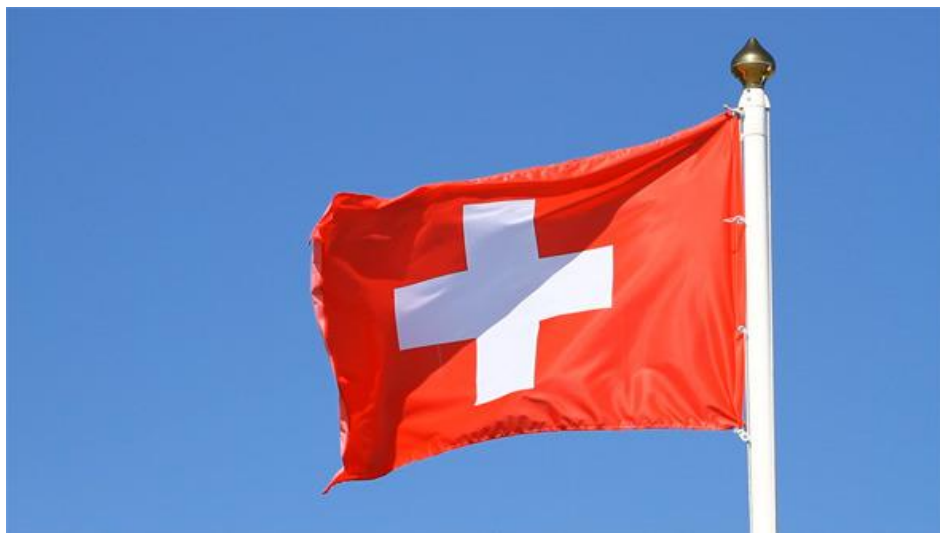
Periodic Table: It's Elementary!



Charles Janet (1926) Left-step Periodic Table

																H	He	1														
																Li	Be	2														
																B	C	N	O	F	Ne	Na	Mg	3								
																Al	Si	P	S	Cl	Ar	K	Ca	4								
														Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	5
														Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba	6
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	Fr	Ra	7
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og		8	

Periodic Table: It's Elementary!



Alfred Werner (1905)
32-column Periodic Table

s																		d										p						
1	H	He																											B	C	N	O	F	Ne
2	Li	Be																											Al	Si	P	S	Cl	Ar
3	Na	Mg																											Ga	Ge	As	Se	Br	Kr
4	K	Ca																	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	In	Sn	Sb	Te	I	Xe
5	Rb	Sr	f																Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
7	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og		

Periodic Table: It's Elementary!



1	H																	He
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg											Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og

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La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Periodic Table: It's Elementary!

IUPAC Periodic Table of the Elements

1 H hydrogen 1.008 [1.0078, 1.0082]																	2 He helium 4.0026
3 Li lithium 6.94 [6.938, 6.997]	4 Be beryllium 9.0122	Key: atomic number Symbol name conventional atomic weight standard atomic weight										13 B boron 10.81 [10.806, 10.821]	14 C carbon 12.011 [12.009, 12.012]	15 N nitrogen 14.007 [14.006, 14.008]	16 O oxygen 15.999 [15.999, 16.000]	17 F fluorine 18.998	18 Ne neon 20.180
11 Na sodium 22.990	12 Mg magnesium 24.305 [24.304, 24.307]	3	4	5	6	7	8	9	10	11	12	13 Al aluminium 26.982	14 Si silicon 28.085 [28.084, 28.086]	15 P phosphorus 30.974	16 S sulfur 32.06 [32.059, 32.076]	17 Cl chlorine 35.45 [35.446, 35.457]	18 Ar argon 39.948
19 K potassium 39.098	20 Ca calcium 40.078(4)	21 Sc scandium 44.956	22 Ti titanium 47.867	23 V vanadium 50.942	24 Cr chromium 51.996	25 Mn manganese 54.938	26 Fe iron 55.845(2)	27 Co cobalt 58.933	28 Ni nickel 58.693	29 Cu copper 63.546(3)	30 Zn zinc 65.38(2)	31 Ga gallium 69.723	32 Ge germanium 72.630(8)	33 As arsenic 74.922	34 Se selenium 78.971(8)	35 Br bromine 79.904 [79.901, 79.907]	36 Kr krypton 83.798(2)
37 Rb rubidium 85.468	38 Sr strontium 87.62	39 Y yttrium 88.906	40 Zr zirconium 91.224(2)	41 Nb niobium 92.906	42 Mo molybdenum 95.95	43 Tc technetium	44 Ru ruthenium 101.07(2)	45 Rh rhodium 102.91	46 Pd palladium 106.42	47 Ag silver 107.87	48 Cd cadmium 112.41	49 In indium 114.82	50 Sn tin 118.71	51 Sb antimony 121.76	52 Te tellurium 127.60(3)	53 I iodine 126.90	54 Xe xenon 131.29
55 Cs caesium 132.91	56 Ba barium 137.33	57-71 lanthanoids	72 Hf hafnium 178.49(2)	73 Ta tantalum 180.95	74 W tungsten 183.84	75 Re rhenium 186.21	76 Os osmium 190.23(3)	77 Ir iridium 192.22	78 Pt platinum 195.08	79 Au gold 196.97	80 Hg mercury 200.59	81 Tl thallium 204.38 [204.38, 204.39]	82 Pb lead 207.2	83 Bi bismuth 208.98	84 Po polonium	85 At astatine	86 Rn radon
87 Fr francium	88 Ra radium	89-103 actinoids	104 Rf rutherfordium	105 Db dubnium	106 Sg seaborgium	107 Bh bohrium	108 Hs hassium	109 Mt meitnerium	110 Ds darmstadtium	111 Rg roentgenium	112 Cn copernicium	113 Nh nihonium	114 Fl flerovium	115 Mc moscovium	116 Lv livermorium	117 Ts tennessine	118 Og oganeson



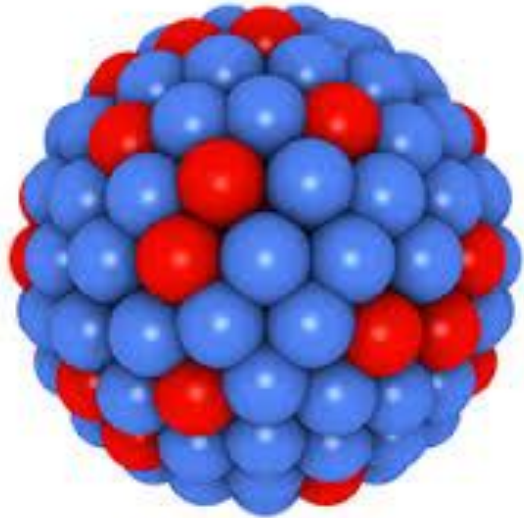
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57 La lanthanum 138.91	58 Ce cerium 140.12	59 Pr praseodymium 140.91	60 Nd neodymium 144.24	61 Pm promethium	62 Sm samarium 150.36(2)	63 Eu europium 151.96	64 Gd gadolinium 157.25(3)	65 Tb terbium 158.93	66 Dy dysprosium 162.50	67 Ho holmium 164.93	68 Er erbium 167.26	69 Tm thulium 168.93	70 Yb ytterbium 173.05	71 Lu lutetium 174.97
89 Ac actinium	90 Th thorium 232.04	91 Pa protactinium 231.04	92 U uranium 238.03	93 Np neptunium	94 Pu plutonium	95 Am americium	96 Cm curium	97 Bk berkelium	98 Cf californium	99 Es einsteinium	100 Fm fermium	101 Md mendelevium	102 No nobelium	103 Lr lawrencium

Periodic Table: It's Elementary!



Chemical Element vs. Elementary Substance

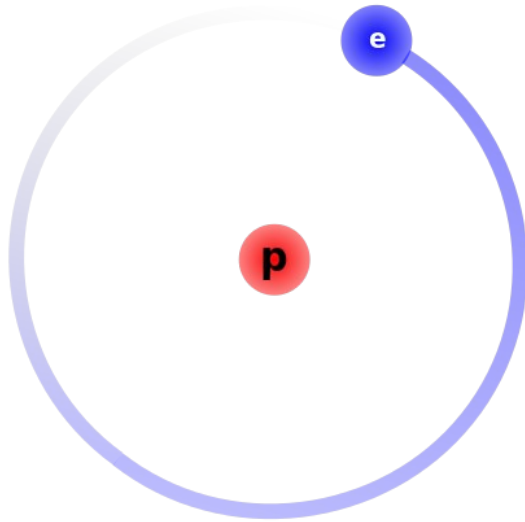


Periodic Table: It's Elementary!

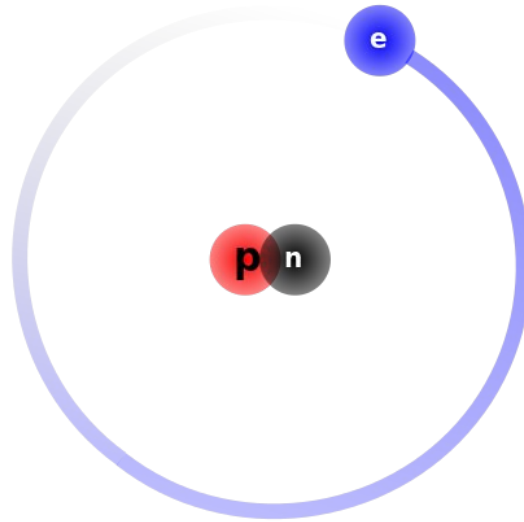


Chemical Element: Species of Equicharged Atomic Nuclei

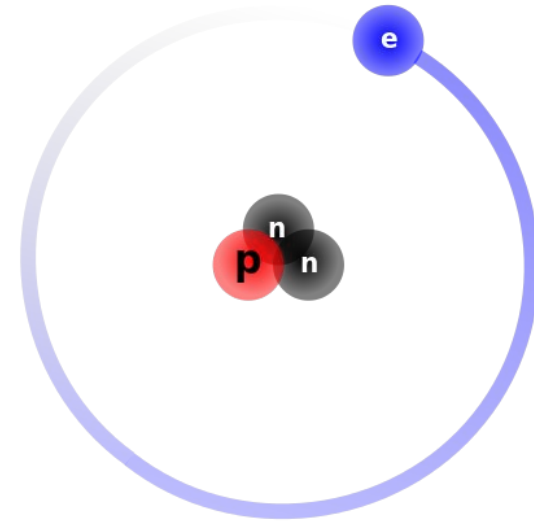
Periodic Table: It's Elementary!



Protium



Deuterium



Tritium

Isotopes of Chemical Element: Chemical Element Representatives with different mass number

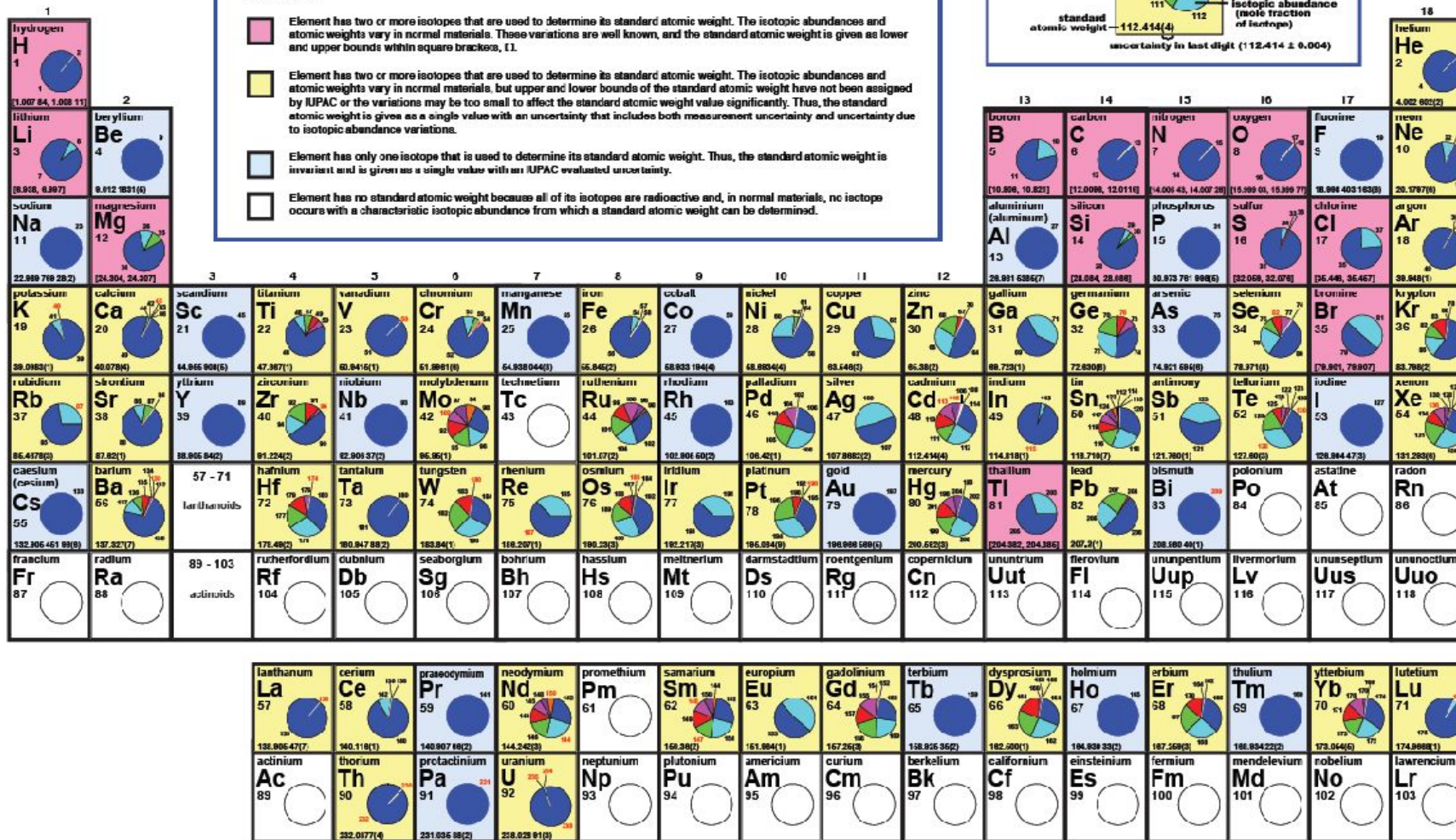
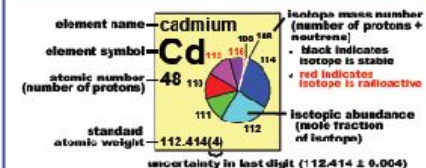
Periodic Table: It's Elementary!

IUPAC Periodic Table of the Isotopes

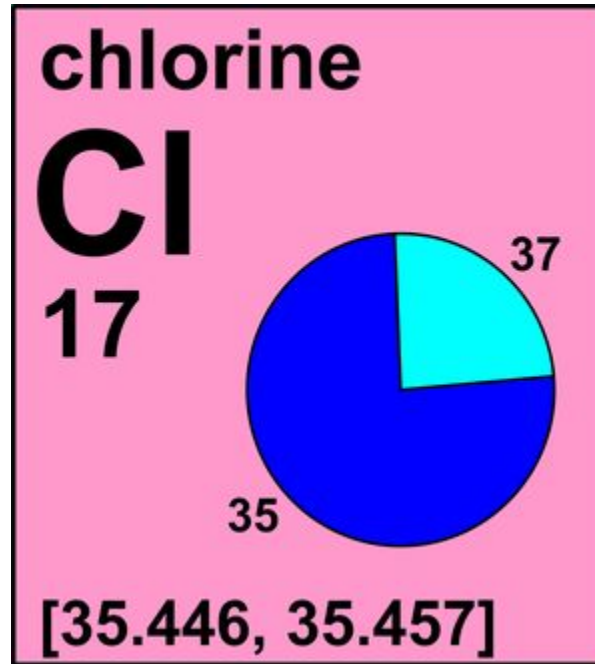
Element Background Color Key

Standard atomic weights are the best estimates by IUPAC of atomic weights that are found in normal materials, which are terrestrial materials that are reasonably possible sources for elements and their compounds in commerce, industry, or science. They are determined using all stable isotopes and selected radioactive isotopes (having relatively long half-lives and characteristic terrestrial isotopic compositions). Isotopes are considered stable (non-radioactive) if evidence for radioactive decay has not been detected experimentally.

- Element has two or more isotopes that are used to determine its standard atomic weight. The isotopic abundances and atomic weights vary in normal materials. These variations are well known, and the standard atomic weight is given as lower and upper bounds within square brackets, [].
- Element has two or more isotopes that are used to determine its standard atomic weight. The isotopic abundances and atomic weights vary in normal materials, but upper and lower bounds of the standard atomic weight have not been assigned by IUPAC or the variations may be too small to affect the standard atomic weight value significantly. Thus, the standard atomic weight is given as a single value with an uncertainty that includes both measurement uncertainty and uncertainty due to isotopic abundance variations.
- Element has only one isotope that is used to determine its standard atomic weight. Thus, the standard atomic weight is invariant and is given as a single value with an IUPAC evaluated uncertainty.
- Element has no standard atomic weight because all of its isotopes are radioactive and, in normal materials, no isotope occurs with a characteristic isotopic abundance from which a standard atomic weight can be determined.



Periodic Table: It's Elementary!

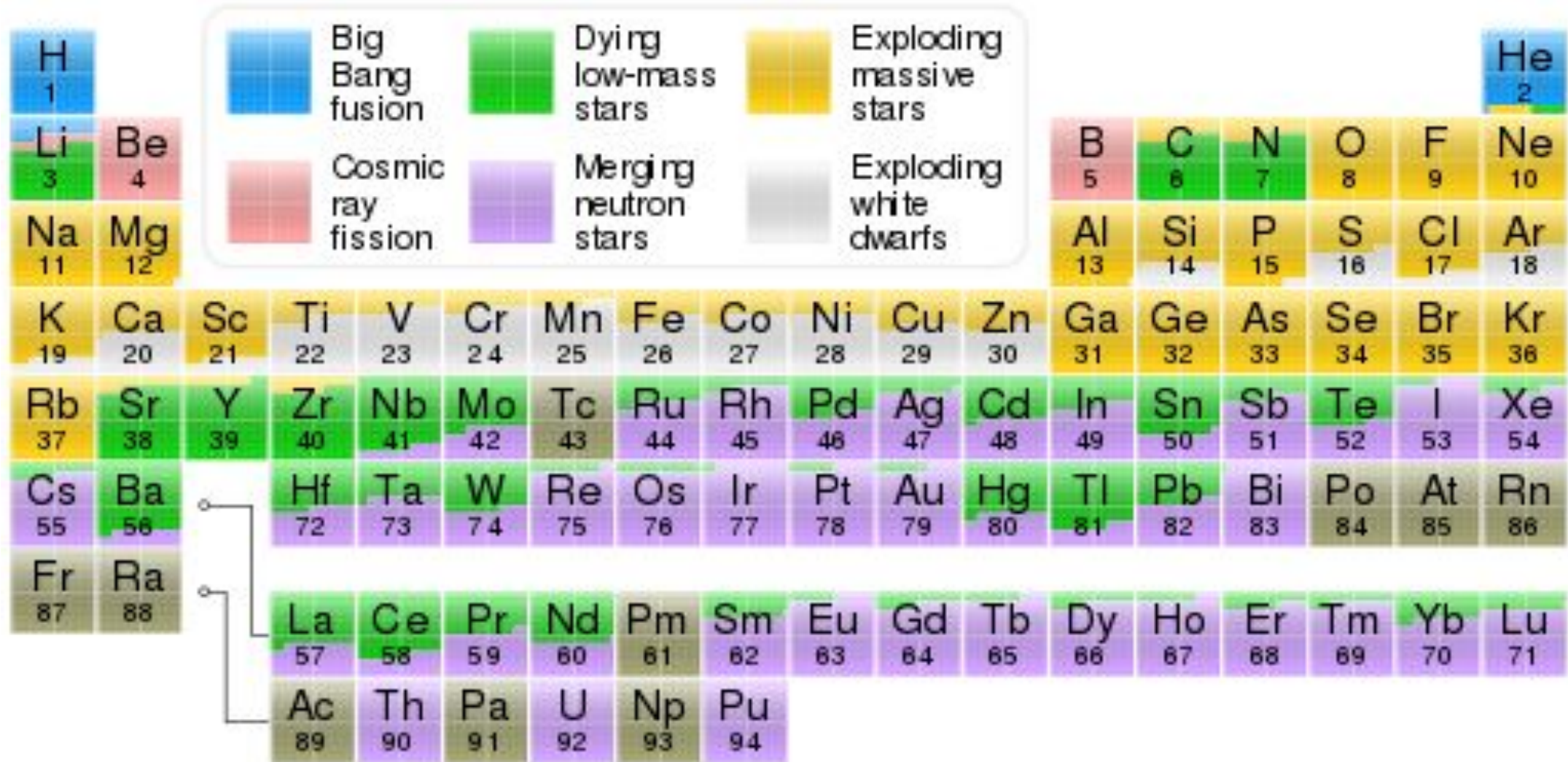


Pie Chart

$$35 \cdot 0.7576 + 37 \cdot 0.2424 = 35.453 \text{ [a.m.u]}$$

Isotopic Abundance

Periodic Table: It's Elementary!



Periodic Table: It's Elementary!

3										2										1						0																										
PRIMORDIAL																																																	1	4		
RADIOGENIC																																																	1	2		
SYNTHETIC																																																	7	9		
																									3	4																										
																									11	12	14	16	19	20	23	24																				
																									5	6	7	8	9	10	11	12																				
																									2.04	2.55	3.04	3.44	3.98	-	0.93	1.31																				
																									27	28	31	32	35.5	40	39	40																				
																									13	14	15	16	17	18	19	20																				
																									1.61	1.90	2.19	2.58	3.16	-	0.82	1.00																				
																									45	48	51	52	55	56	59	59	64	65	70	73	75	79	80	84	85.5	88										
																									21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38										
																									1.36	1.54	1.63	1.66	1.55	1.83	1.88	1.91	1.90	1.65	1.81	2.01	2.18	2.55	2.96	-	0.82	0.95										
																									89	91	93	96	98	101	103	106	108	112	115	119	122	128	127	131	133	137										
																									39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56										
																									1.22	1.33	1.6	2.16	2.10	2.2	2.28	2.20	1.93	1.69	1.78	1.96	2.05	2.1	2.66	2.60	0.79	0.89										
139	140	141	144	145	150	152	157	159	162.5	165	167	169	173	175	178.5	181	184	186	190	192	195	197	201	204	207	209	209	210	222	223	226																					
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	Fr	Ra																					
57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88																					
1.10	1.12	1.13	1.14	-	1.17	-	1.20	-	1.22	1.23	1.24	1.25	-	1.0	1.3	1.5	1.7	1.9	2.2	2.2	2.2	2.4	1.9	1.8	1.8	1.9	2.0	2.2	-	0.7	0.9																					
227	232	231	238	237	244																																															
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og																							
89	90	91	92	93	94																																															
1.1	1.3	1.5	1.7	1.3	1.3																																															

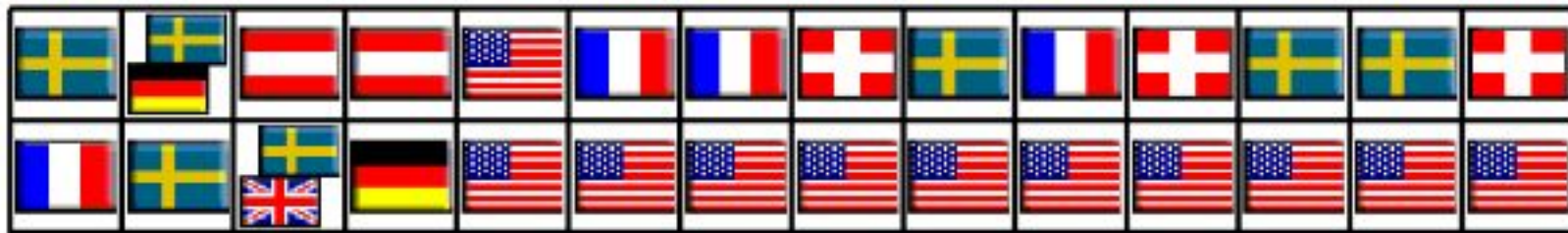
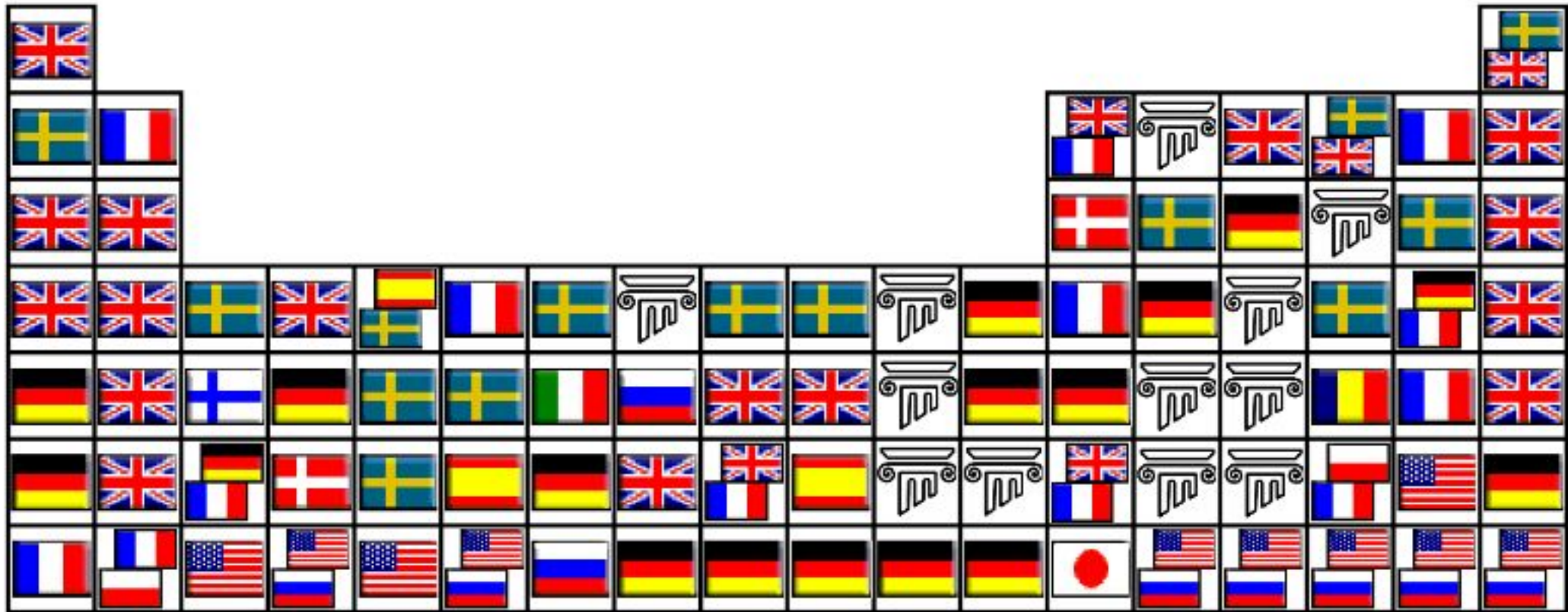
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3										2										1						0				
ATMOPHILES										LITHOPHILES										CHALCOPHILES						SIDEROPHILES		1	4	1
																												H	He	
																												1	2	2
																												Li	Be	
																												7	9	3
																												Na	Mg	
																												11	12	3
																												B	C	
																												5	6	4
																												Al	Si	
																												27	28	4
																												Al	Si	
																												45	48	5
																												Sc	Ti	
																												21	22	5
																												Sc	Ti	
																												89	91	6
																												Y	Zr	
																												39	40	7
																												Y	Zr	
																												139	140	7
																												La	Ce	
																												227	232	8
																												Ac	Th	

Crust 5-70 km thick	Mantle 2,867 km thick	Outer core 2,256 km thick	Inner core 1,216 km thick
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Victor Goldschmidt Geochemical Classification

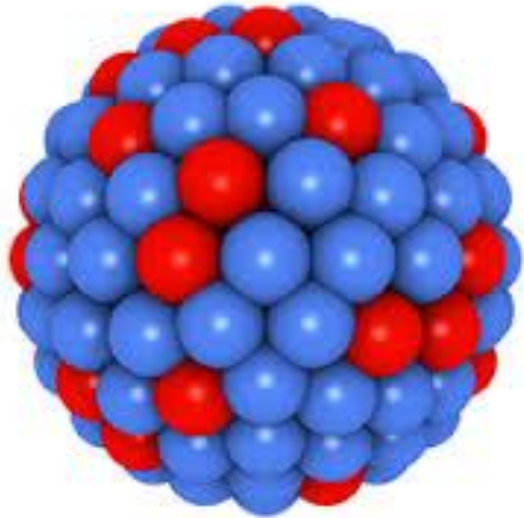
Periodic Table: It's Elementary!



Periodic Table: It's Elementary!



Chemical Element vs. Elementary Substance



Periodic Table: It's Elementary!

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37 Rb rubidium 85.468	38 Sr strontium 87.62	39 Y yttrium 88.906	40 Zr zirconium 91.224(2)	41 Nb niobium 92.906	42 Mo molybdenum 95.95	43 Tc technetium	44 Ru ruthenium 101.07(2)	45 Rh rhodium 102.91	46 Pd palladium 106.42	47 Ag silver 107.87	48 Cd cadmium 112.41	49 In indium 114.82	50 Sn tin 118.71	51 Sb antimony 121.76	52 Te tellurium 127.60(3)	53 I iodine 126.90	54 Xe xenon 131.29
55 Cs caesium 132.91	56 Ba barium 137.33	57-71 lanthanoids	72 Hf hafnium 178.49(2)	73 Ta tantalum 180.95	74 W tungsten 183.84	75 Re rhenium 186.21	76 Os osmium 190.23(3)	77 Ir iridium 192.22	78 Pt platinum 195.08	79 Au gold 196.97	80 Hg mercury 200.59	81 Tl thallium 204.38 [204.38, 204.39]	82 Pb lead 207.2	83 Bi bismuth 208.98	84 Po polonium	85 At astatine	86 Rn radon
87 Fr francium	88 Ra radium	89-103 actinoids	104 Rf rutherfordium	105 Db dubnium	106 Sg seaborgium	107 Bh bohrium	108 Hs hassium	109 Mt meitnerium	110 Ds darmstadtium	111 Rg roentgenium	112 Cn copernicium	113 Nh nihonium	114 Fl flerovium	115 Mc moscovium	116 Lv livermorium	117 Ts tennessine	118 Og oganeson



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57 La lanthanum 138.91	58 Ce cerium 140.12	59 Pr praseodymium 140.91	60 Nd neodymium 144.24	61 Pm promethium	62 Sm samarium 150.36(2)	63 Eu europium 151.96	64 Gd gadolinium 157.25(3)	65 Tb terbium 158.93	66 Dy dysprosium 162.50	67 Ho holmium 164.93	68 Er erbium 167.26	69 Tm thulium 168.93	70 Yb ytterbium 173.05	71 Lu lutetium 174.97
89 Ac actinium	90 Th thorium 232.04	91 Pa protactinium 231.04	92 U uranium 238.03	93 Np neptunium	94 Pu plutonium	95 Am americium	96 Cm curium	97 Bk berkelium	98 Cf californium	99 Es einsteinium	100 Fm fermium	101 Md mendelevium	102 No nobelium	103 Lr lawrencium

Periodic Table: It's Elementary!

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