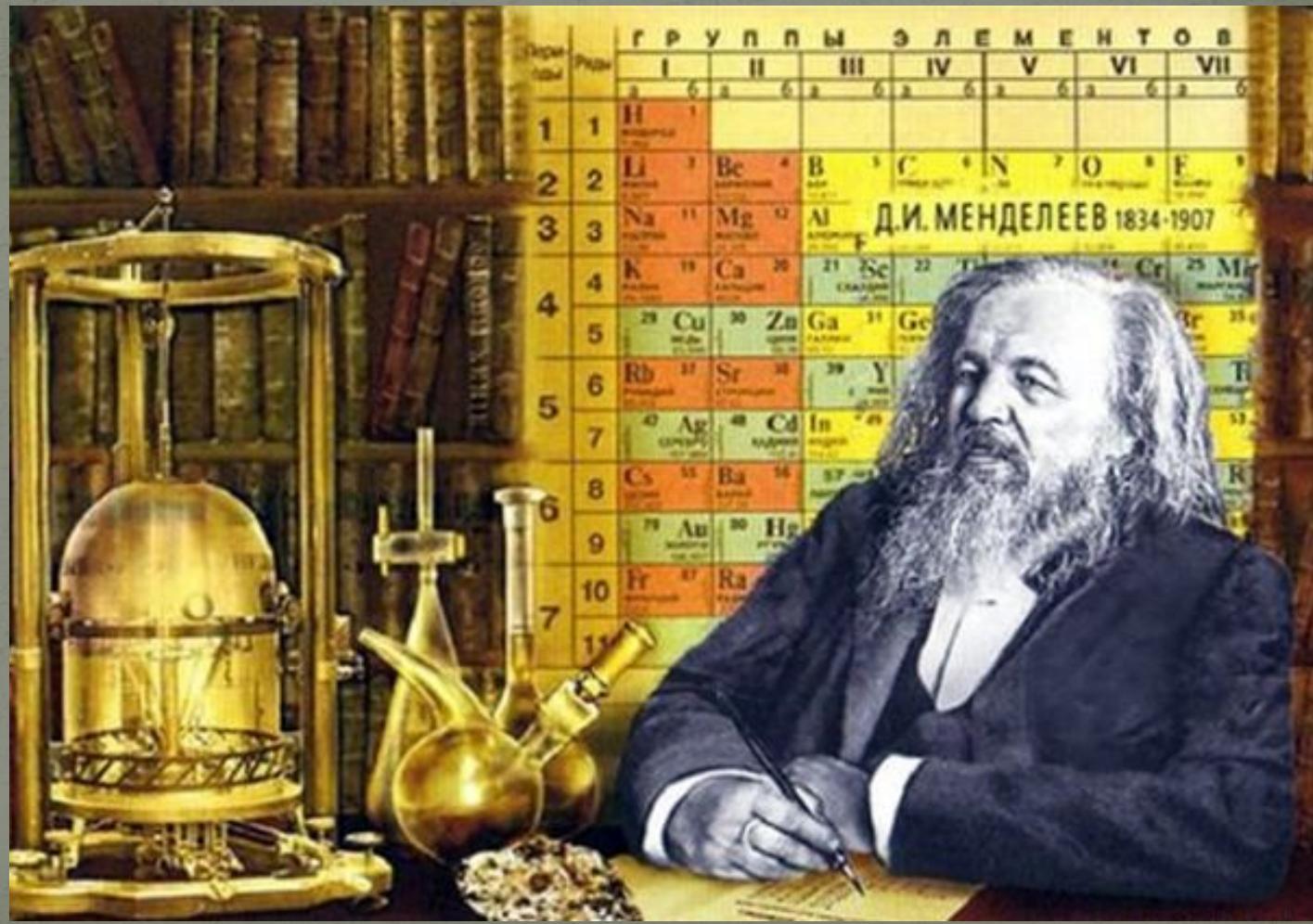


Dmitri Ivanovich Mendeleev



Dmitry Ivanovich Mendeleev was born on 27 January 1834 in Tobolsk. Mendeleev's parents were Maria Mendeleeva (Kornilieva) and Ivan Mendeleev.



*Мария Дмитриевна Менделеева
(урожденная Корнильева)
(1793-1850) –
мать Д.И.Менделеева*

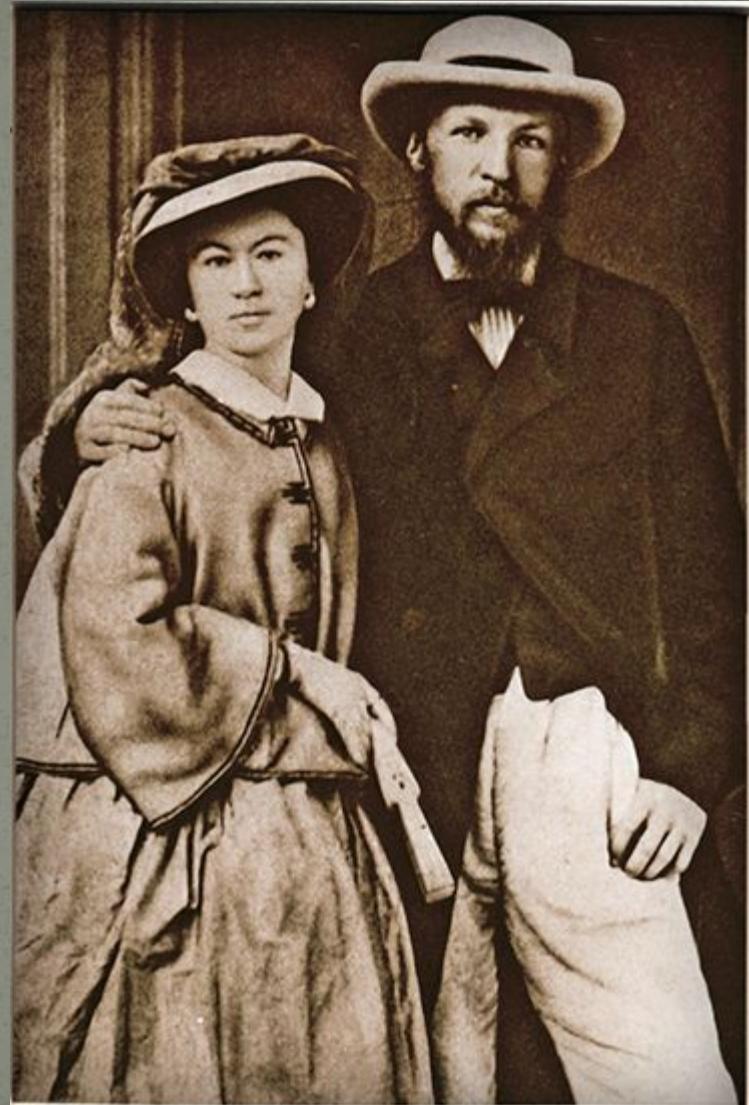


*Иван Павлович Менделеев
(1783 – 1847)
отец Д.И. Менделеева*



In 1849 his family moved to Saint Petersburg. In 1850 Mendeleev joined The Main Pedagogical Institute. Following the graduation he developed tuberculosis and was forced to relocate to the Crimean Peninsula. Living there, Mendeleev became a science master of the Simferopol gymnasium. In 1857 after recovery he arrived in Saint Petersburg.

From 1859 to 1861 Mendeleev worked in Heidelberg. In April 1862 he married Feozva Nikitichna Leshcheva. Mendeleev obtained a lot of awards from different scientific organizations but he resigned from Saint Petersburg University in 1890.



Periodic Law was opened by Mendeleev in 1869.

<p style="text-align: right;">Ostwald, <u>Die Elemente</u>, S. 63 ff,</p> <p style="text-align: center;"><u>Periodic System</u></p> <p style="text-align: center;">of elements according to their electrochemical properties.</p> <p style="text-align: center;">D. Mendeleev.</p>	
	$I = 18$ $R = 90$ $? = 100$ $I = 51$ $N = 94$ $I = 182$ $C = 50$ $I = 46$ $W = 116$ $H = 55$ $R = 119$ $H = 115$ $S = 36$ $R = 129$ $D = 114$ $K = 37$ $R = 126$ $C = 119$
$H = 1$ Li $I = 11$ $C = 8$ $S = 18$ $O = 16$ $F = 11$ $N = 15$	$? = 8$ $I = 22$ $C = 63$ $I = 21$ $I = 65$ $I = 24$ $I = 61$ $I = 31$ $I = 50$ $I = 31$ $I = 55$ $I = 32$ $I = 53$ $I = 35$ $I = 50$ $I = 39$ $R = 131$ $I = 46$ $? = 45$ $I = 58$ $I = 60$ $I = 67$
	$I = 101$ $I = 102$ $R = 116$ $R = 115$ $I = 118$ $S = 112$ $R = 910$ $I = 118$ $I = 112$ $R = 113$ $R = 101$ $I = 112$ $R = 117$ $R = 117$
	$I = 100$ $I = 102$ $R = 115$ $R = 117$
<p style="margin: 0;">Essai d'une système des éléments d'après leurs poids atomiques et fonctions chimiques de Mendeleff.</p> <p style="margin: 0;">Par Jules Duhem, Université de Paris.</p>	
<p style="margin: 0;">Malgrès l'admission de la théorie atomique, Mendeleff a conservé les éléments dans leur ordre de croissance de leur poids atomique, sans déduire aucun.</p>	
$18 \frac{II}{I} 69.$	<p style="margin: 0;">Tous ces éléments sont dans l'ordre de croissance de leur poids atomique, mais déduits.</p>

In 1906 the Nobel Committee for Chemistry suggested to the Swedish Academy to award the Nobel Prize in Chemistry for 1906 to Mendeleev for his discovery of the periodic system. But Mendeleev's candidacy was rejected.

		I	Периодическая система химических элементов Д. И. Менделеева						VII	VIII		
1	1	H							(H)	He		
1	1	H	1 1,00794 водород									
2	2	Li	3	Be	4	B	6	C	7	N	8	O
3	3	Na	11	Mg	12	Al	14	Si	15	P	16	S
4	4	K	19	Ca	20	Sc	21	Ti	22	V	23	Cr
5	5	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se
6	6	Rb	37	Sr	38	Y	39	Zr	40	Nb	41	Mo
7	7	Ag	47	Cd	48	In	50	Sn	51	Sb	52	Te
8	8	Cs	55	Ba	56	La*	57	Hf	72	Ta	73	W
9	9	Au	79	Hg	80	Tl	81	Pb	82	Bi	83	Po
10	10	Fr	87	Ra	88	Ac**	89	Rf	104	Db	105	Sg
11	11	Rg	[272]	Uub	[285]	(Ut)	[1]	Uuq	[287]	(Up)	[1]	Uuh



Периодический закон открыт
Д.И. Менделеевым в 1869 г.

* Лантаноиды

Ce 58	Pr 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	No 67	Er 68	Tm 69	Yb 70	Lu 71
140,12 церий	140,9077 презеодим	144,24 неодим	[145] прометий	130,36 самарий	151,96 европий	157,25 гадолиний	158,9254 тербий	162,50 дилютрозий	164,9304 гольмий	167,26 эрбий	168,9342 тулий	173,04 иттербий	174,967 лютений

** Актиноиды

Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102	Lr 103
232,0381 торий	[231] протактиний	238,0289 уран	[237] нейптиний	[244] плутоний	[243] америций	[247] киюрий	[247] берклий	[251] калифорний	[252] эйнштейний	[253] фермий	[257] менделеевий	[259] нобелий	[260] лоуренсий

Целое число в скобках – массовое число наиболее устойчивого изотопа

Dmitri Mendeleev died of influenza in Saint Petersburg in 1907.



ИСТОЧНИКИ

https://www.google.ru/search?q=менделеев&newwindow=1&biw=1368&bih=632&source=lnms&tbs=isch&sa=X&ved=oahUKEwikiOVrtXPAhVIESwKHeLzDrwQ_AUIBigB&dpr=1

[https://www.google.ru/search?q=менделеев&newwindow=1&biw=1368&bih=632&source=lnms&tbs=isch&sa=X&ved=oahUKEwiQ5PX12NXPAhXG2SwKHbQPBCwQ_AUIBigB#imgrc=j4dgTkXoc4_YrM%3A](https://www.google.ru/search?q=менделеев&newwindow=1&biw=1368&bih=632&source=lnms&tbs=isch&sa=X&ved=oahUKEwikiOVrtXPAhVIESwKHeLzDrwQ_AUIBigB&dpr=1#newwindow=1&tbs=isch&q=%Do%BC%Do%B5%Do%BD%Do%B4%Do%B5%Do%BB%Do%B5%Do%B5%Do%B5%Do%B2+%Do%B2+%Do%BC%Do%BE%Do%BB%Do%BE%Do%B4%Do%BE%D1%81%D1%82%Do%B8&imgrc=VoBI8wFpITg-4M%3A)

https://www.google.ru/search?q=самая+первая+таблица+менделеева&newwindow=1&biw=1368&bih=632&tbs=isch&imgil=gs72tcWR7M5cJM%253A%253B8u4QtRu83xsn_M%253Bht tp%25253A%25252F%25252Fmetkere.com%25252F2009%25252Fo8%25252Ftable.html&source=iu&pf=m&fir=gs72tcWR7M5cJM%253A%252C8u4QtRu83xsn_M%252C_&usg=__69GAWq TMNo7As1fyxUIUt4Qqd5k%3D&ved=oahUKEwii8Kvc2tXPAhVFYJoKHegtCu4QyjcIKQ&ei=z2f-V-LMMsXA6QTo26jwDg#imgdii=gs72tcWR7M5cJM%3A%3Bgs72tcWR7M5cJM%3A%3BS OZgnkpMBhVIIM%3A&imgrc=gs72tcWR7M5cJM%3A