Practical pharmacology Part 1

pharma ology It is sience of the drugs

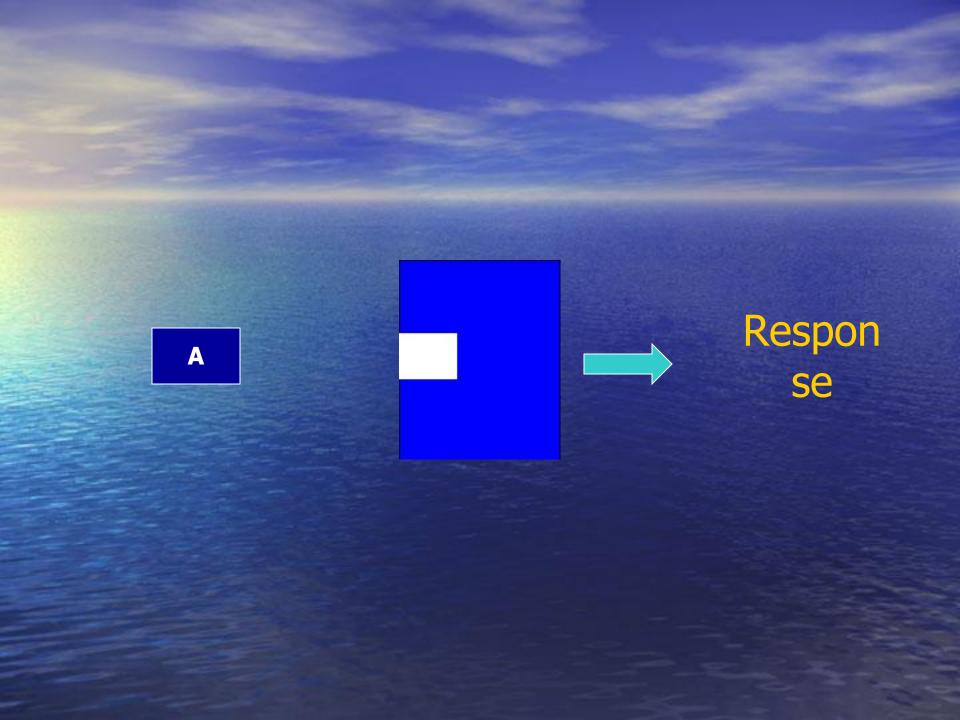
Drug

Science

2What is Days It is the chemical that affect physiological body function through interaction with receptors

?What is Drug

It is the chemical that affect physiological body function through interaction with receptors



Covalent bonds

Ionic bonds

Hydrogen bonds

Vander waals bonds

Covalent bonds

Ionic bonds

Hydrogen bonds

Vander waals bonds

Covalent bonds

Strong

irreversible

Alkylating agents

Covalent bonds

Ionic bonds

Hydrogen bonds

Vander waals bonds

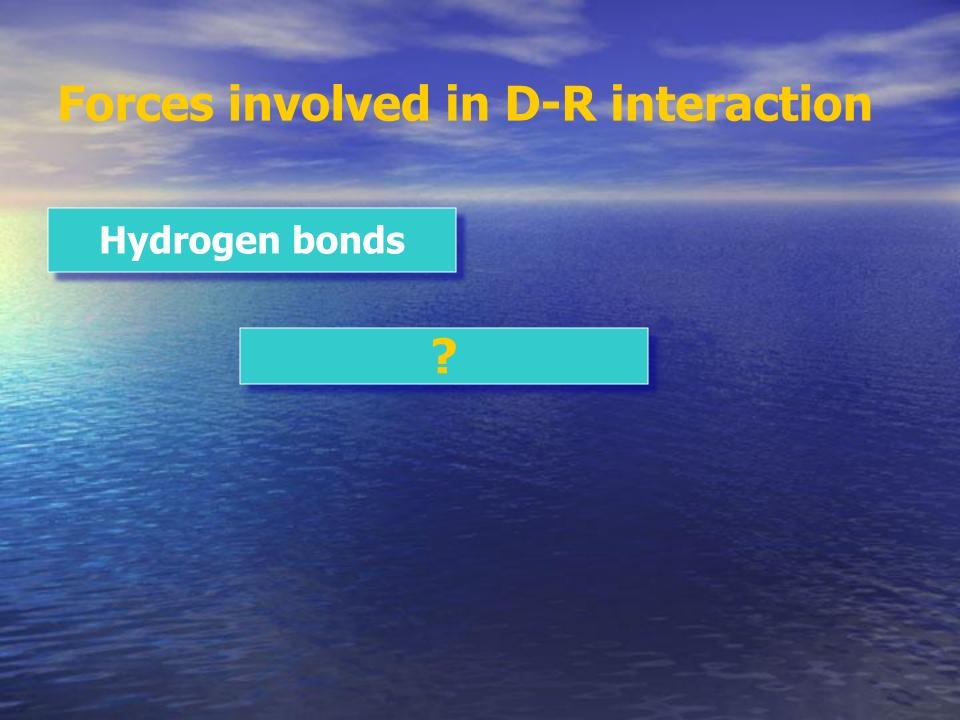
Forces involved in D-R interaction **Ionic bonds** common Affected by pH

Covalent bonds

Ionic bonds

Hydrogen bonds

Vander waals bonds

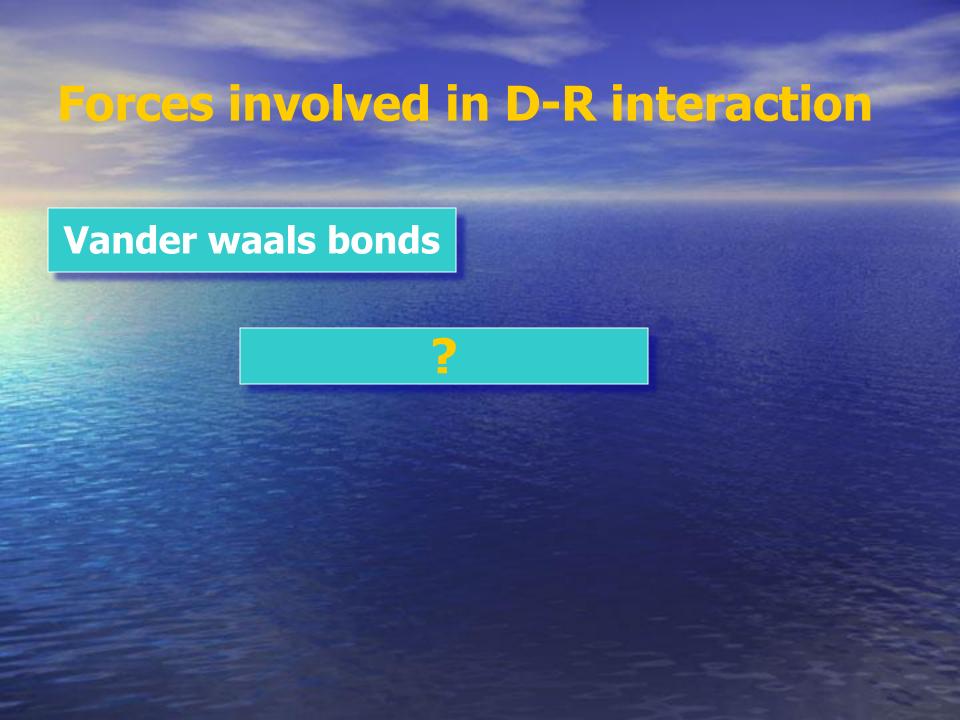


Covalent bonds

Ionic bonds

Hydrogen bonds

Vander waals bonds



Receptors types

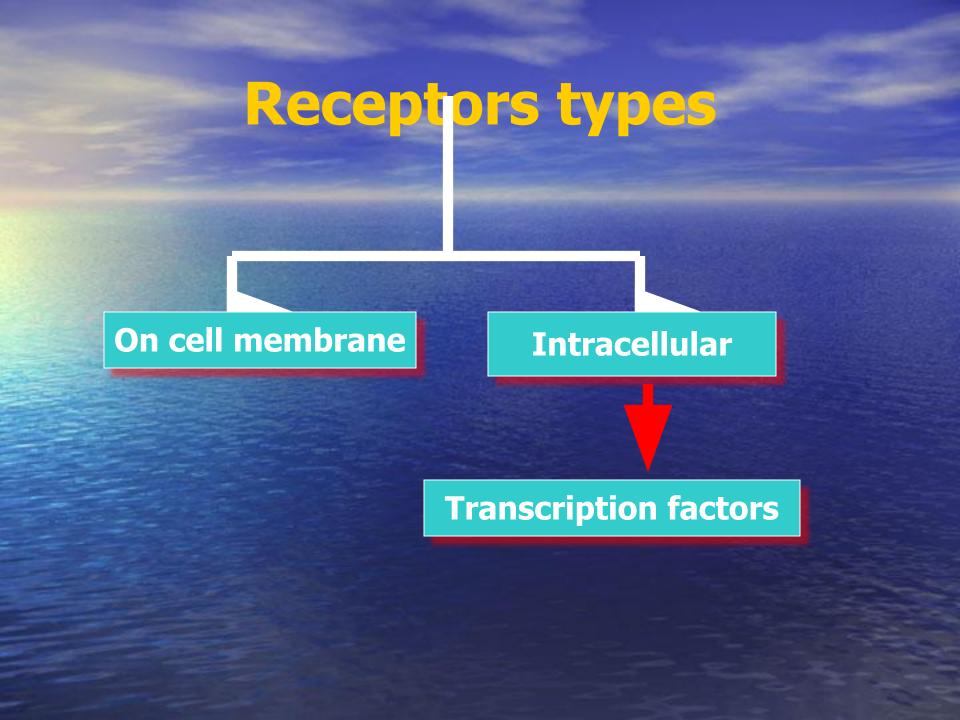
On cell membrane

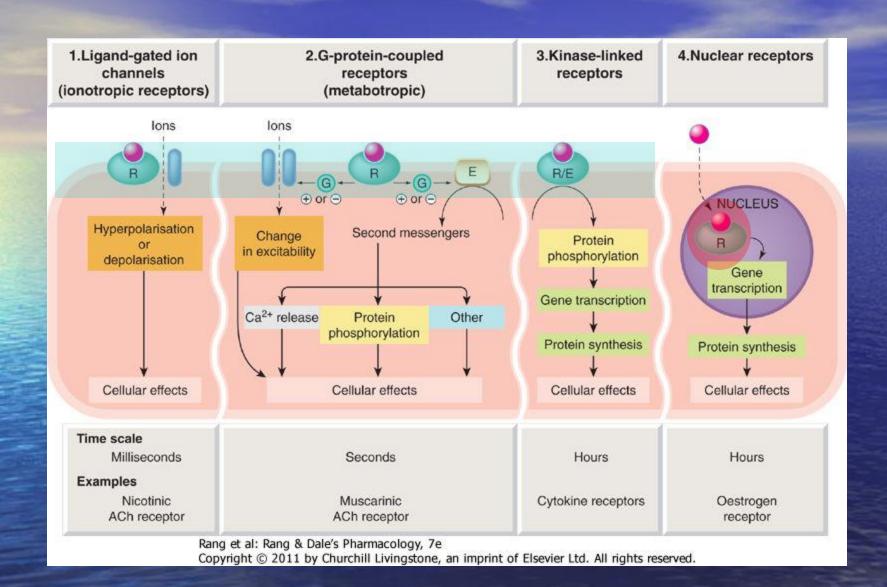
Intracellular

GPCR

Receptor with intrinsic ion channel

Enzyme linked receptors







PR LDR

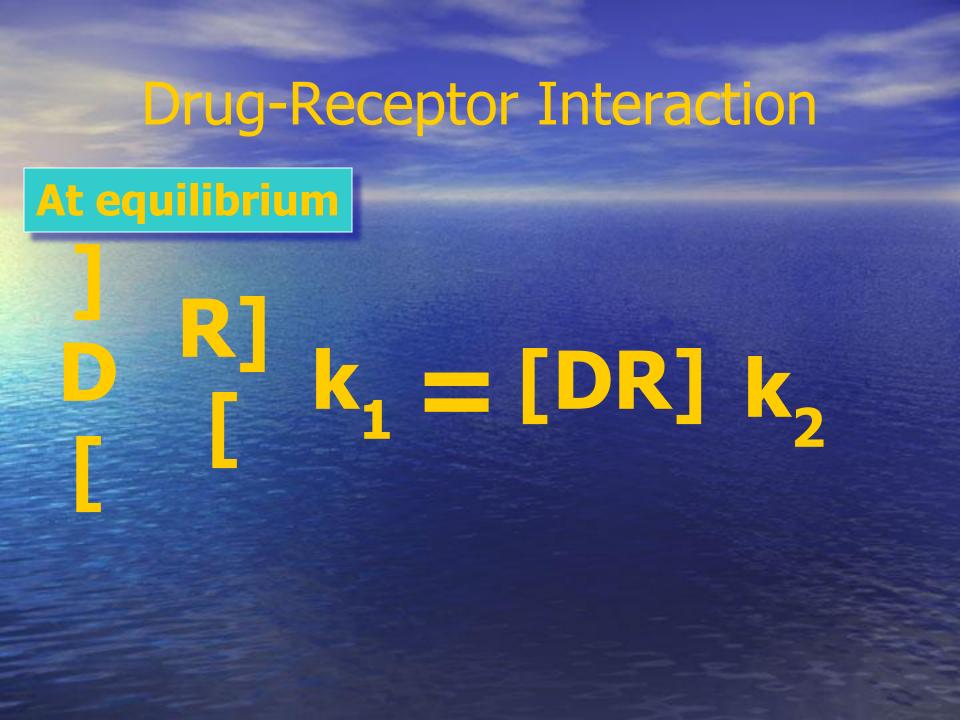
K₁ is association rate constant

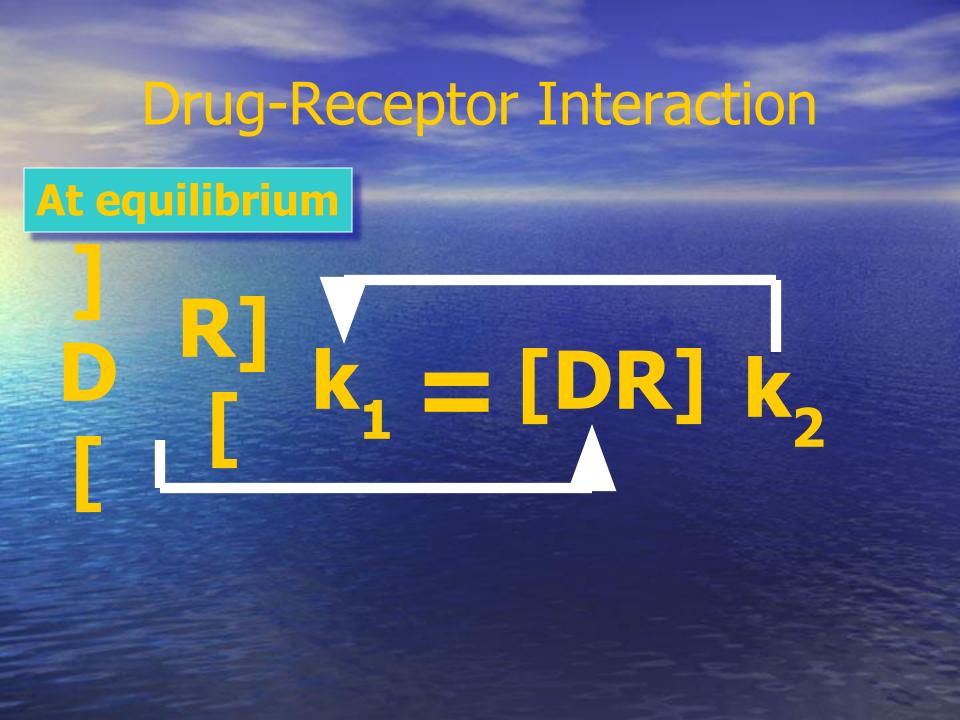
Drug-Receptor Interaction Compared to the second s

DR 2 DH R

K₂ is dissociation rate constant

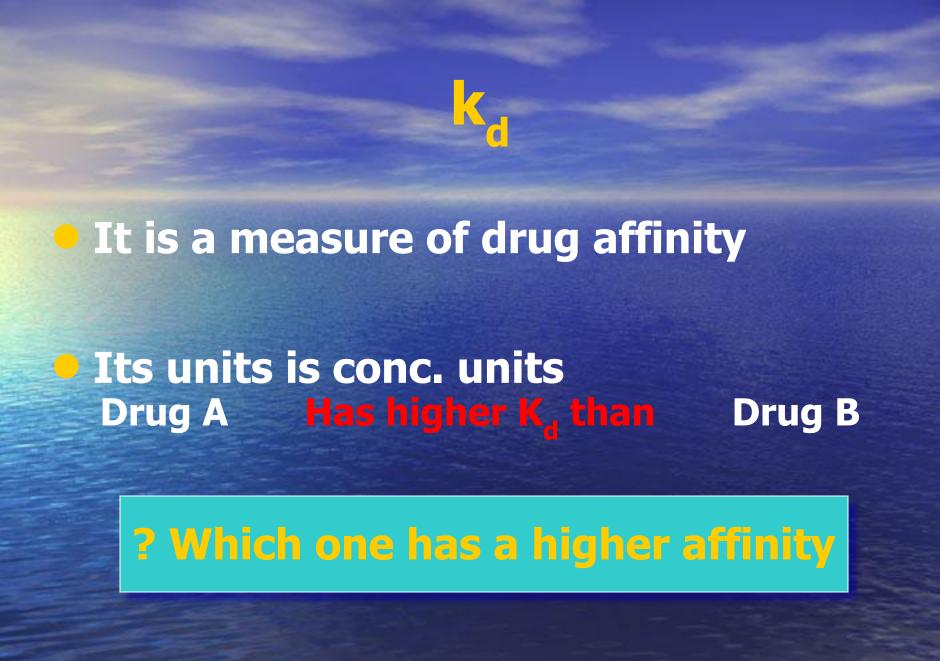
Drug-Recepto Interaction At equilibrium





Drug-Receptor Interaction

Kd (dissociation equilibrium constant) is conc. of the drug that bind 50 % of the receptors





Affinity

The ability of the drug to bind to the receptor

Measured by Kd

Both agonist and antagonist have affinity to their receptors

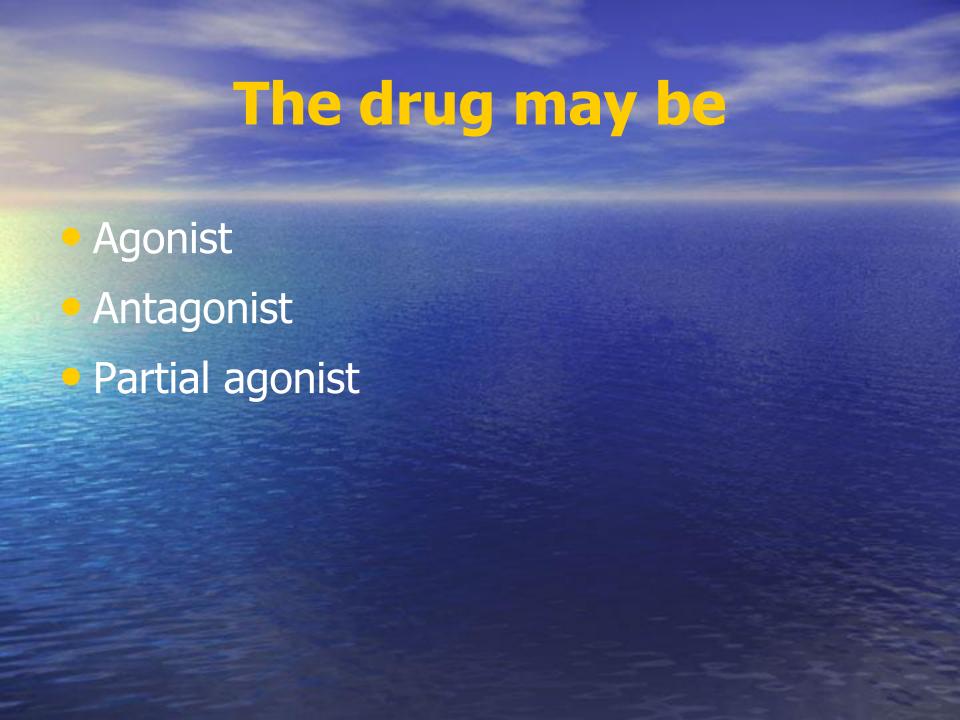
Efficacy

 It is the ability of the drugs to elicit pharmacological effect

Measured by E_{max}

Agonist has efficacy and antagonist has no efficacy

potency The ability of the drug to produce response at lower conc. Measured by ED50



Agonist Has affinity and efficacy **□ IA=1**

Antagonist

Has affinity but no efficacy

• IA=0

Partial agonist

Has affinity and efficacy

● IA=0-1

antagonism Physical Chemical Physiological Pharmacokinetic

pharmacodynamic

Chemical antagonism

- One drug reacts chemically with an active drug to form an inactive compound,
- It involves precipitation, complexation, neutralization redox reaction.
- Intended □treatment of heavy metal toxicity by complexation with chelators.
- Incidental

 complexatin of tetracycline calcium in dairy products.

Physiological antagonism

- 2 drugs act on different sites in the same or different system.
- a- Intended ☐ Norepinephrine in case of anaphylaxis.
- b-Incidental

 patient taking barbiturates for anxiety, co-administration of anti-tussive (ephedrine).

Pharmacokinetic antagonism (ADME)

- a- Intended
 forced alkaline diuresis in management of barb Poisoning.
- b- Incidental

 Barb. + other drugs

Induction of the metabolism of concomitant drugs, their decrease plasma level.

Pharmacodynamic antagonist

Competitive

Non-competitive

Reversible

Surmountable antagonism

Ach. + atropine

Pharmacodynamic antagonist

Competitive

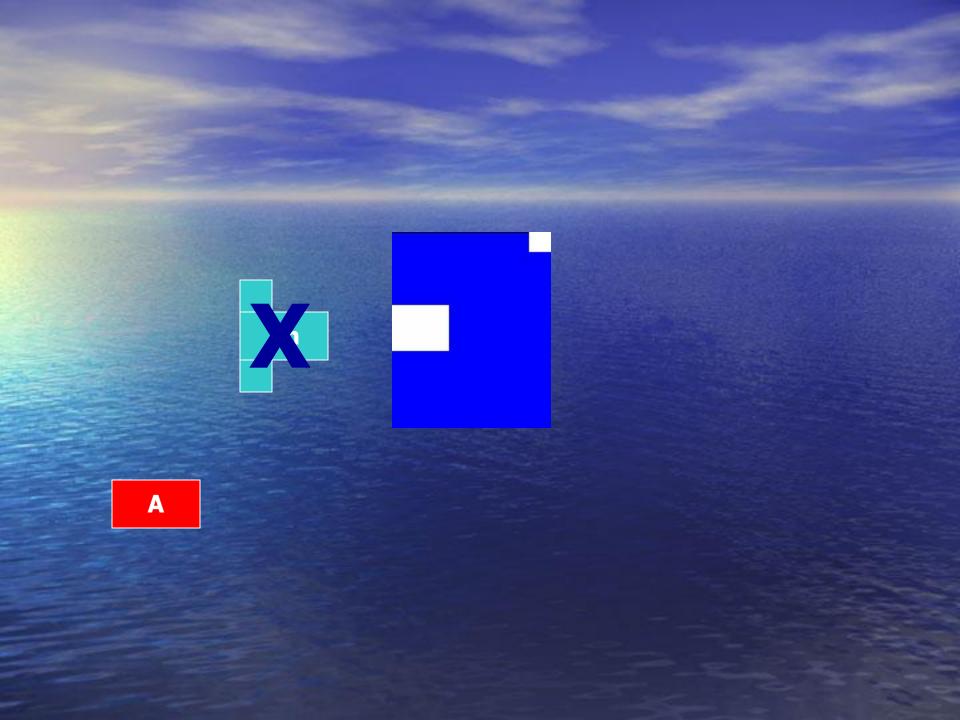
Non-competitive

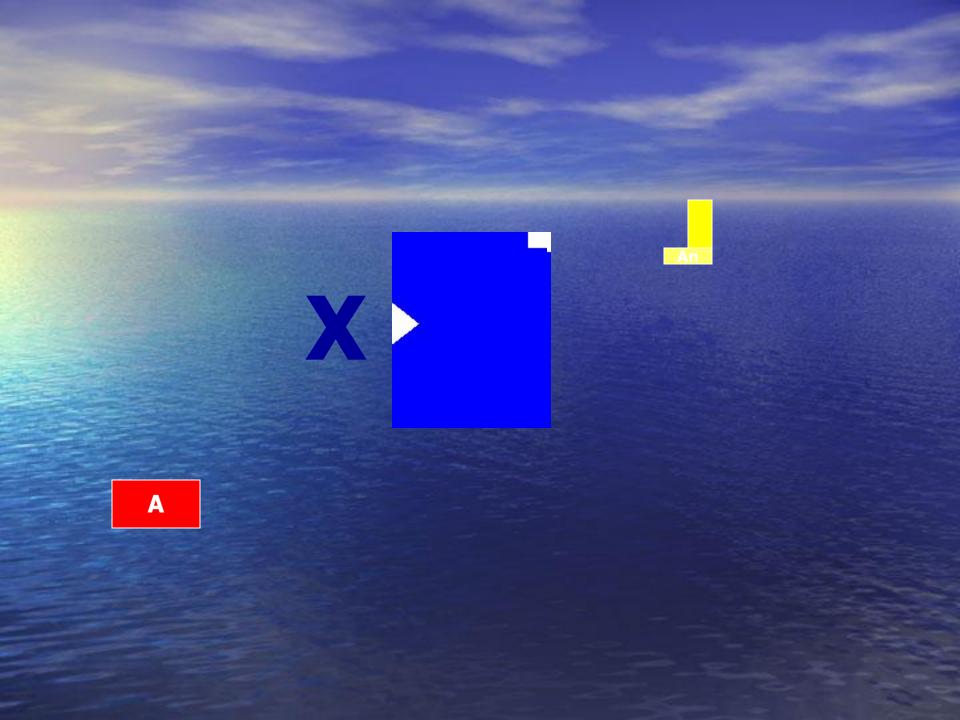
Reversible

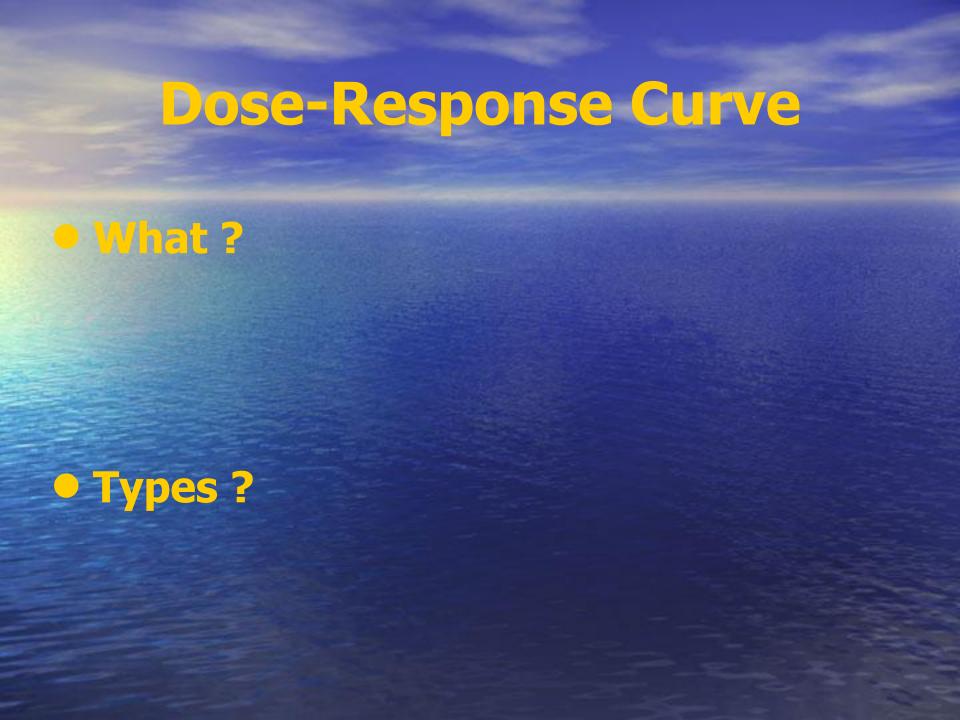
Irreversible

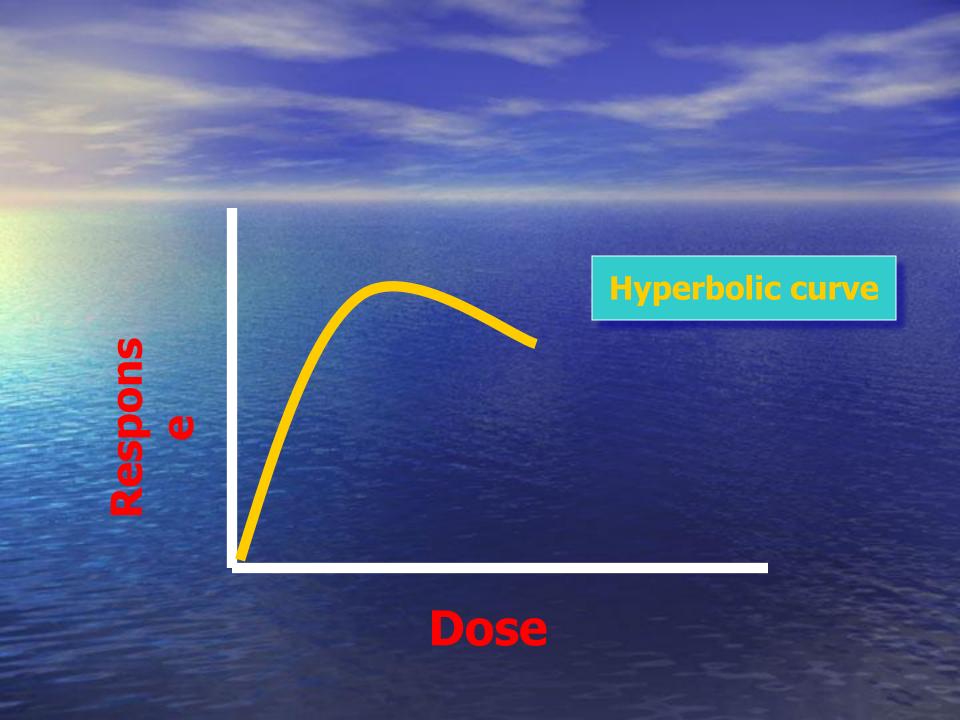
Non-surmountable antagonism

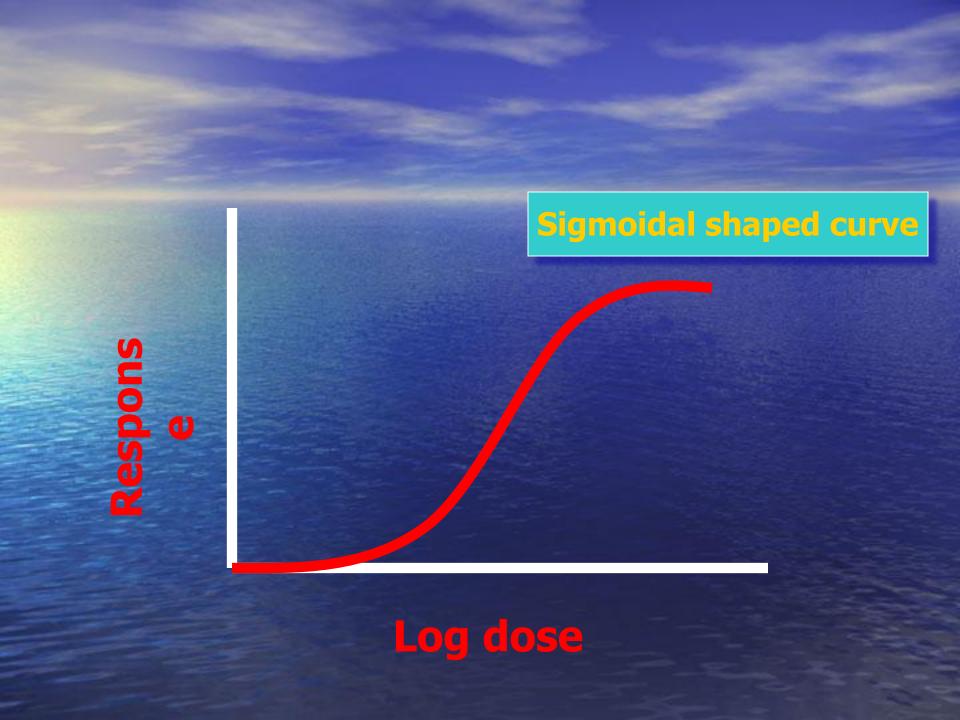
Ach. + succinylcholine

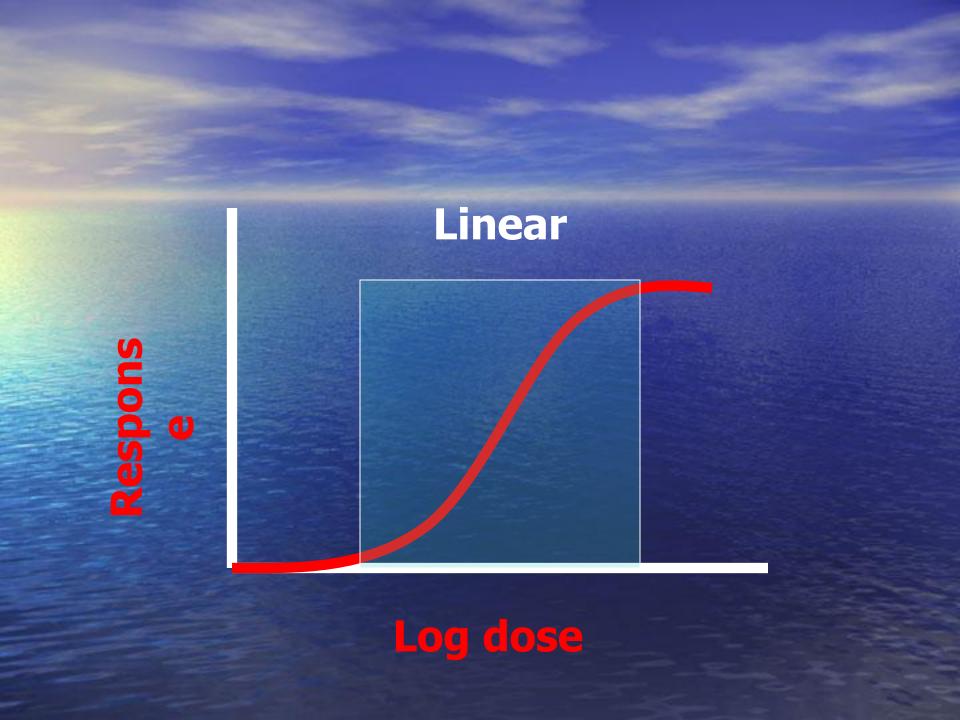


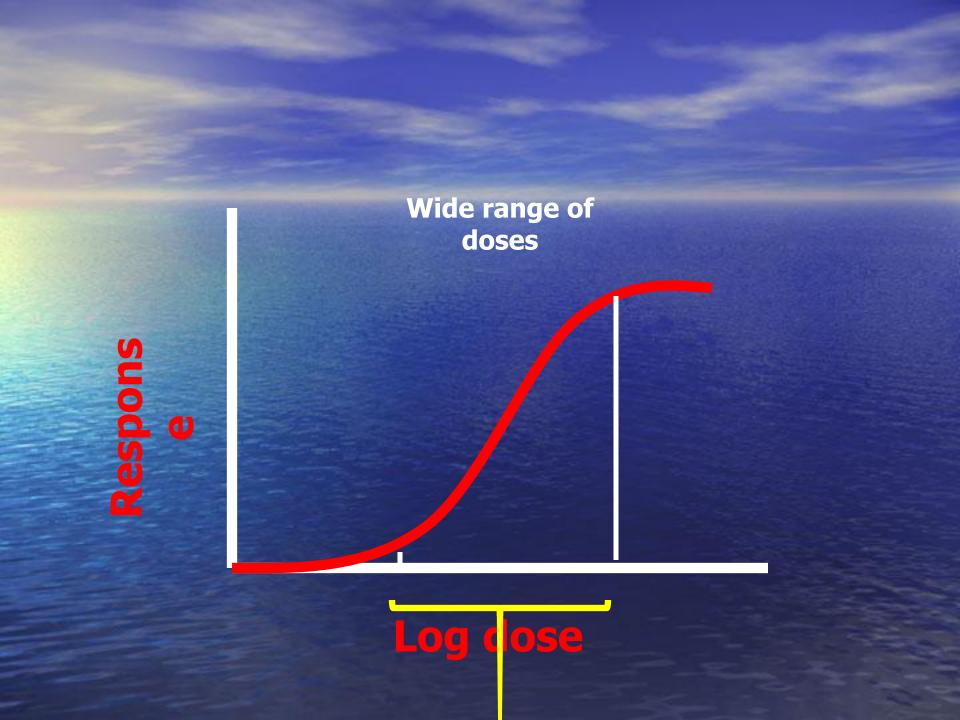


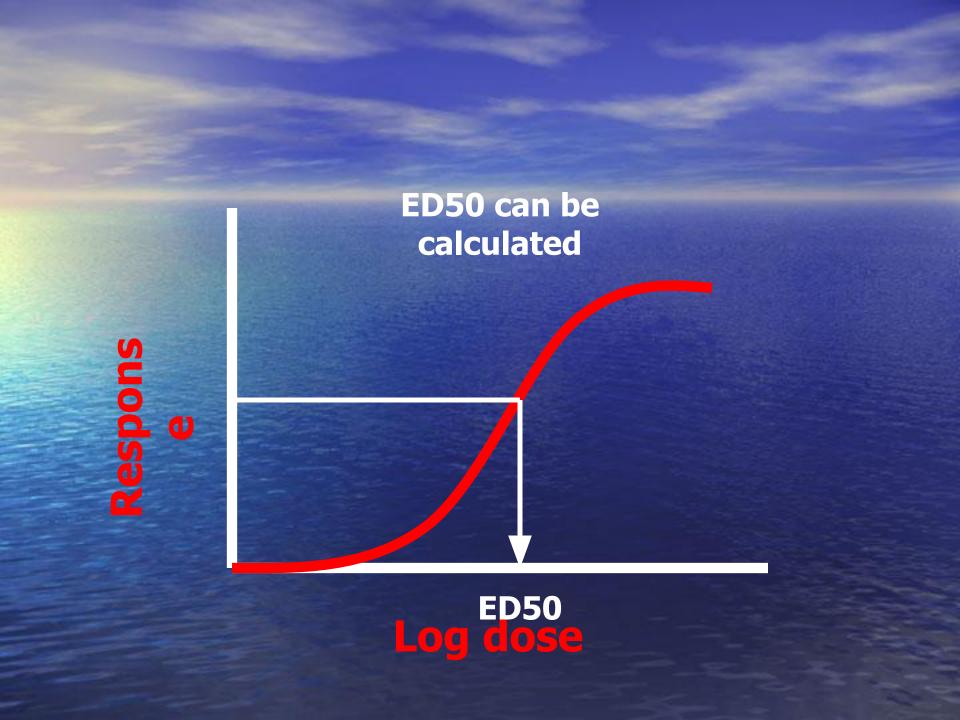










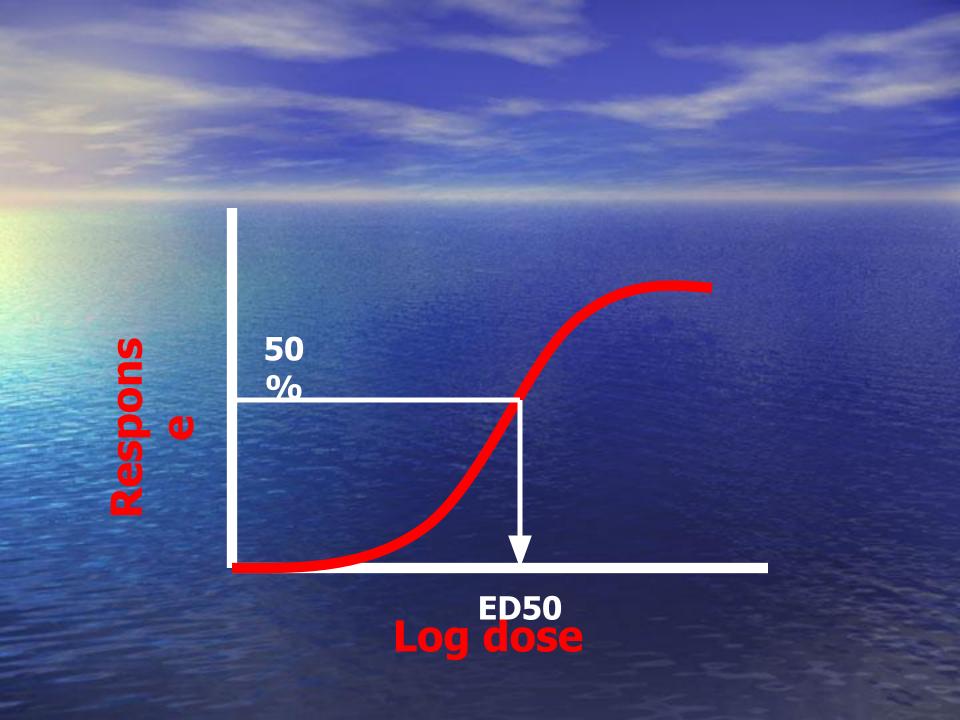


Graded DRC

Depends on graded response

ED50?

The dose that give 50% of maximal response

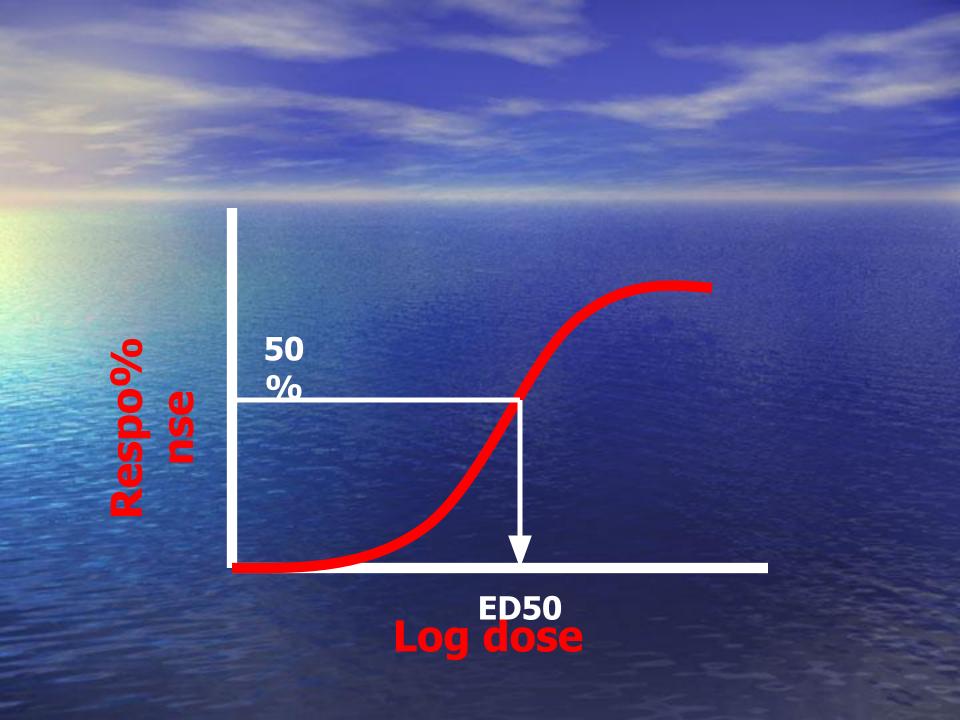


Quantal DRC

Depends on quantal response

ED50 ?

The dose that give response in 50% of population



ED50

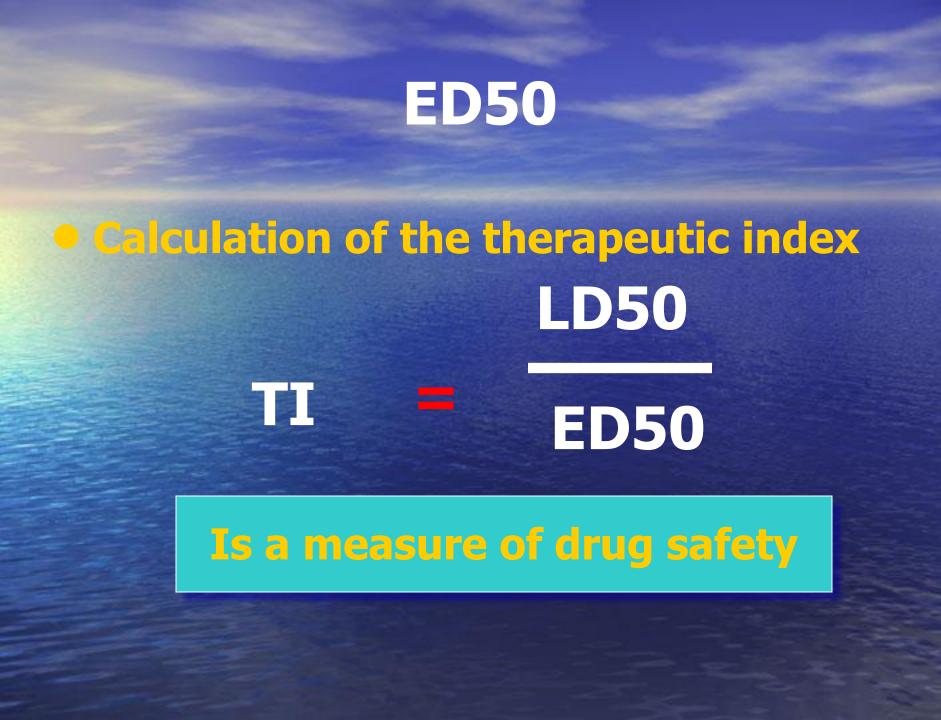
Compare between potencies of two drugs

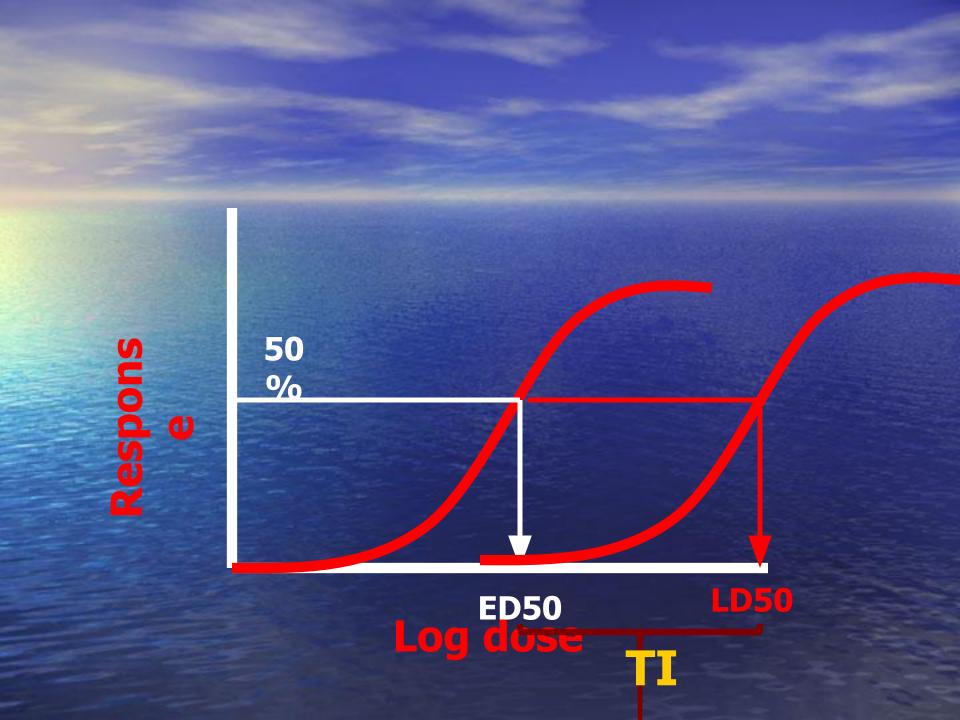
Drug A

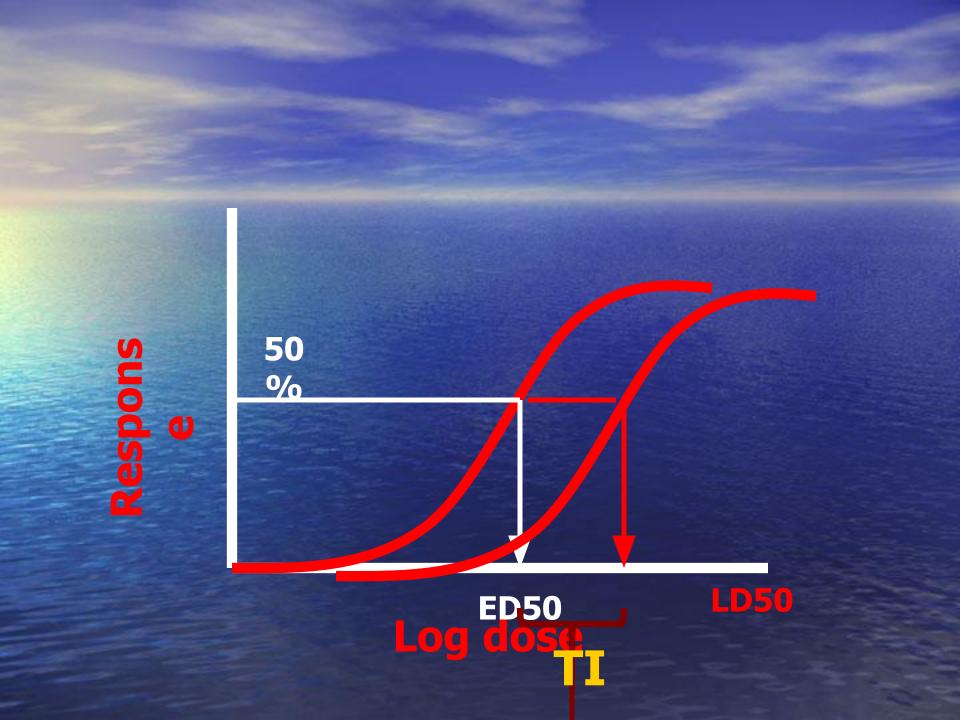
Has higher ED50 than

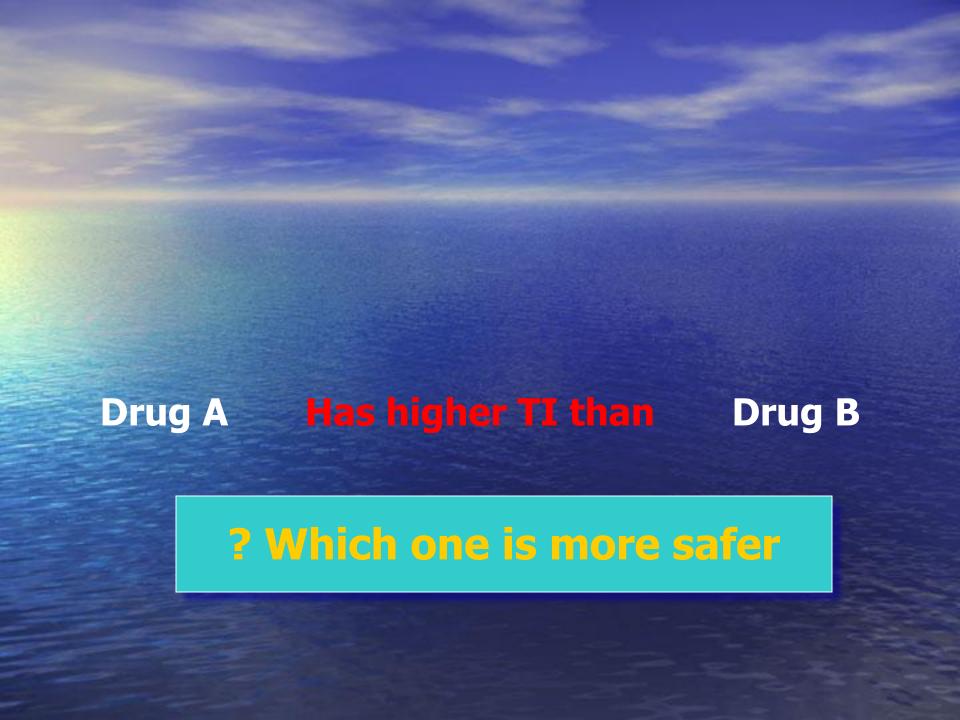
Drug B

? Which one is more potent





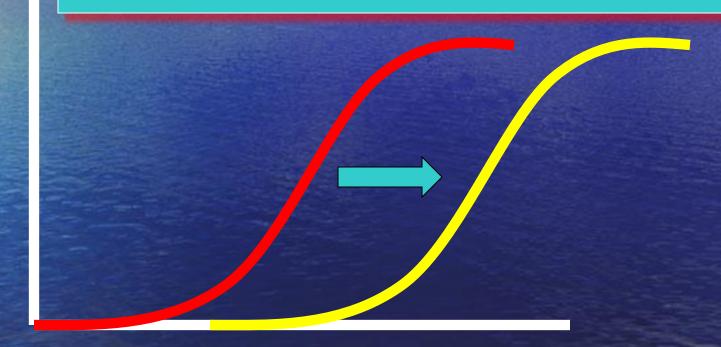




DRC & antagonists

Competitive reversible antagonist

Respons e

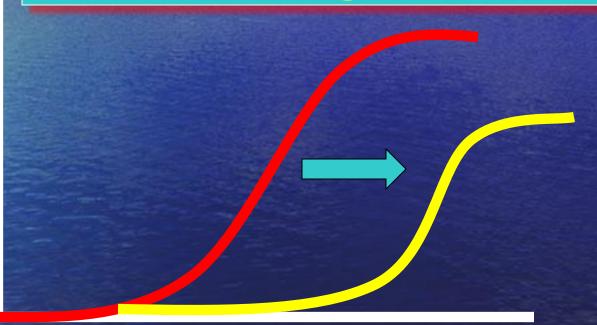


Log dose

DRC & antagonists

Competitive irreversible antagonist

Respons e

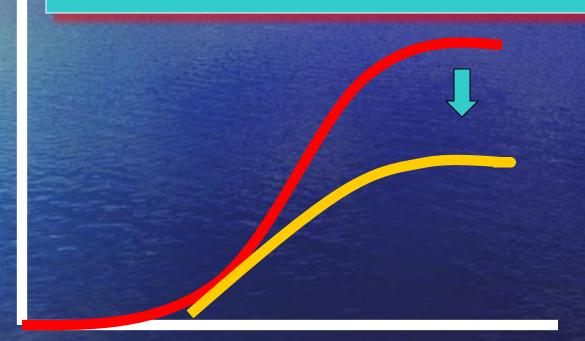


Log dose

DRC & antagonists

non-competitive antagonist

Respons e



Log dose

